

# Synthesis and Optical Properties of $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{Eu}^{2+}$

A. Katelnikovas<sup>a,b</sup>, T. Jüstel<sup>b</sup>, A. Kareiva<sup>a</sup>



<sup>a</sup> Department of General and Inorganic Chemistry, Vilnius University, Naugarduko 24, LT-03225 Vilnius, Lithuania

<sup>b</sup> Department of Chemical Engineering, Münster University of Applied Sciences, Stegerwaldstr. 39, D-48565 Steinfurt, Germany



## Introduction

There are no doubts already that future lighting belongs to solid-state light sources, i.e. white LEDs. The invention of efficient blue LED enabled scientists to produce white light as a combination of blue and yellow components. But such light sources, however, possess high color temperatures and low color rendering indexes. The other possible way for producing white light is based on application of near-UV LED and several phosphors emitting in blue, green and red, therefore the demand of new phosphors emitting in those spectral regions is rapidly increasing.

We demonstrate, that  $\text{Eu}^{2+}$  doped  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}$  shows emission in cyan-green spectral region. The optical properties of synthesized phosphors are discussed as a function of  $\text{Eu}^{2+}$  concentration.

## Experimental Part

All samples were prepared by 2 step solid state reaction from oxides and carbonates. First, starting materials were annealed at  $1400^\circ\text{C}$  for 5 hours under CO atmosphere in order to obtain the desired phase and partially reduce  $\text{Eu}^{3+}$  to  $\text{Eu}^{2+}$ . The complete reduction of europium ions to  $\text{Eu}^{2+}$  was obtained by annealing target samples under 5%  $\text{H}_2$  / 95%  $\text{N}_2$  gas flow for 2 hours in the tube furnace.

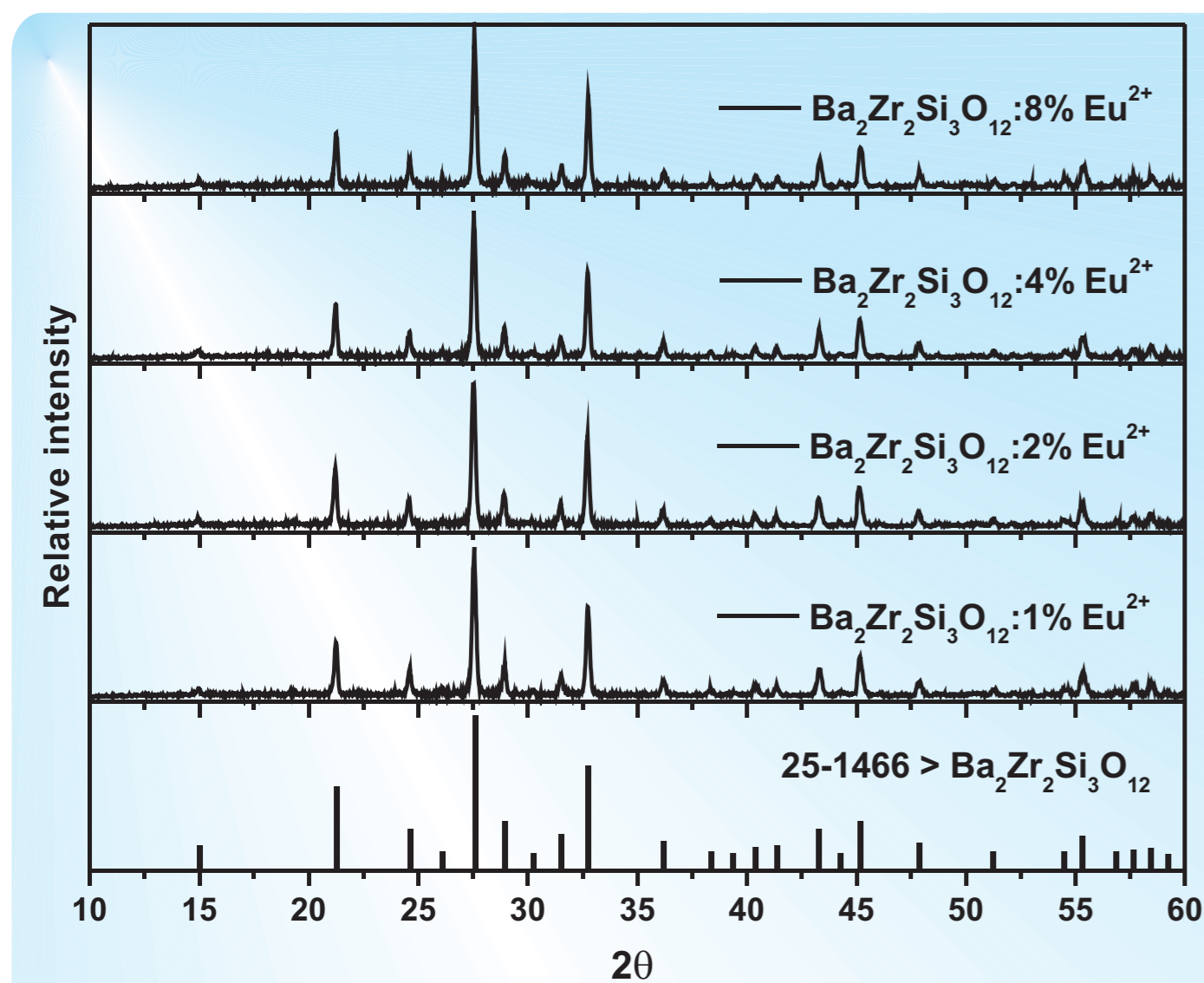


Fig. 1. XRD patterns of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}$  samples with various  $\text{Eu}^{2+}$  concentrations.

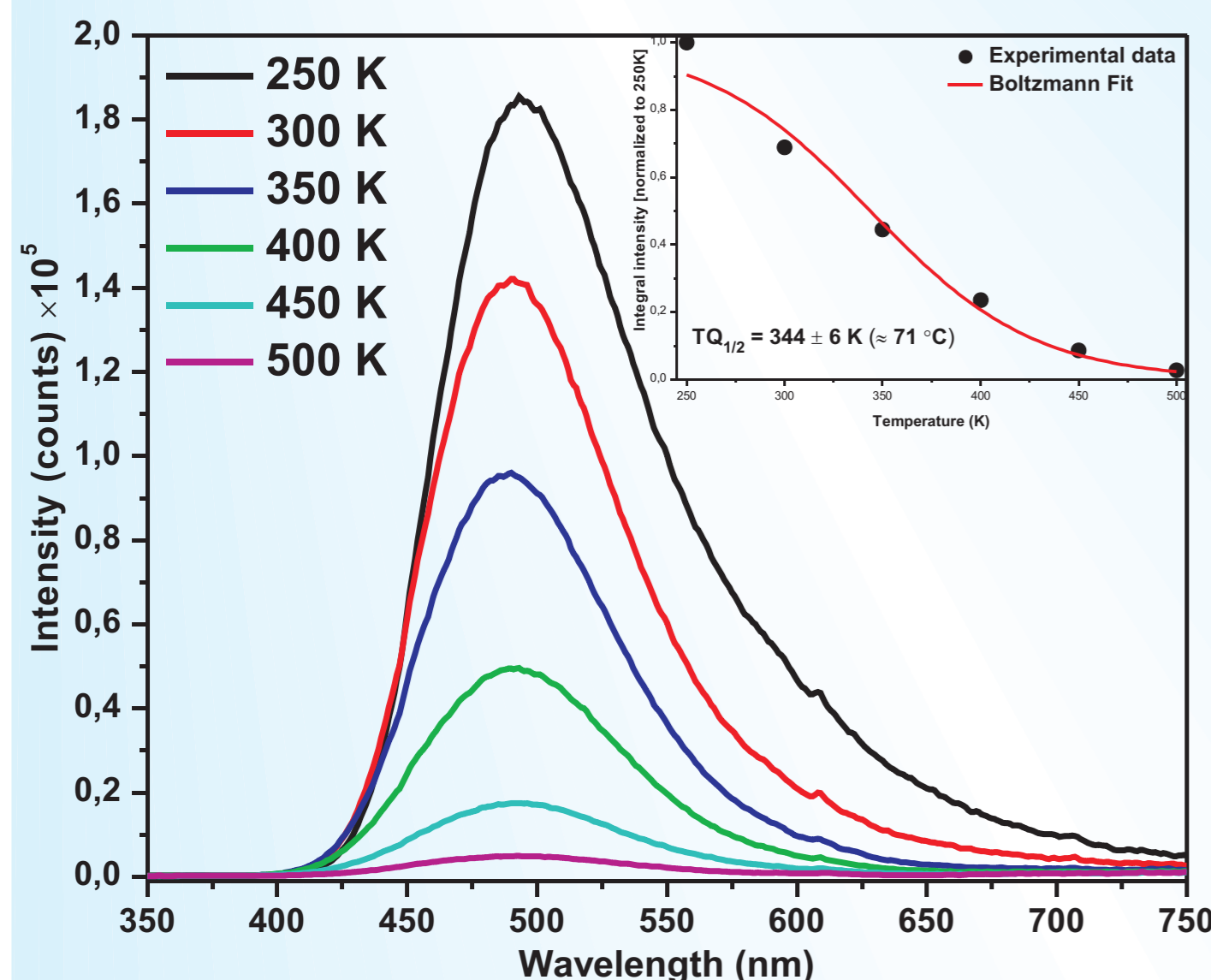


Fig. 4. Temperature dependent emission spectra of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{2\%Eu}^{2+}$ .

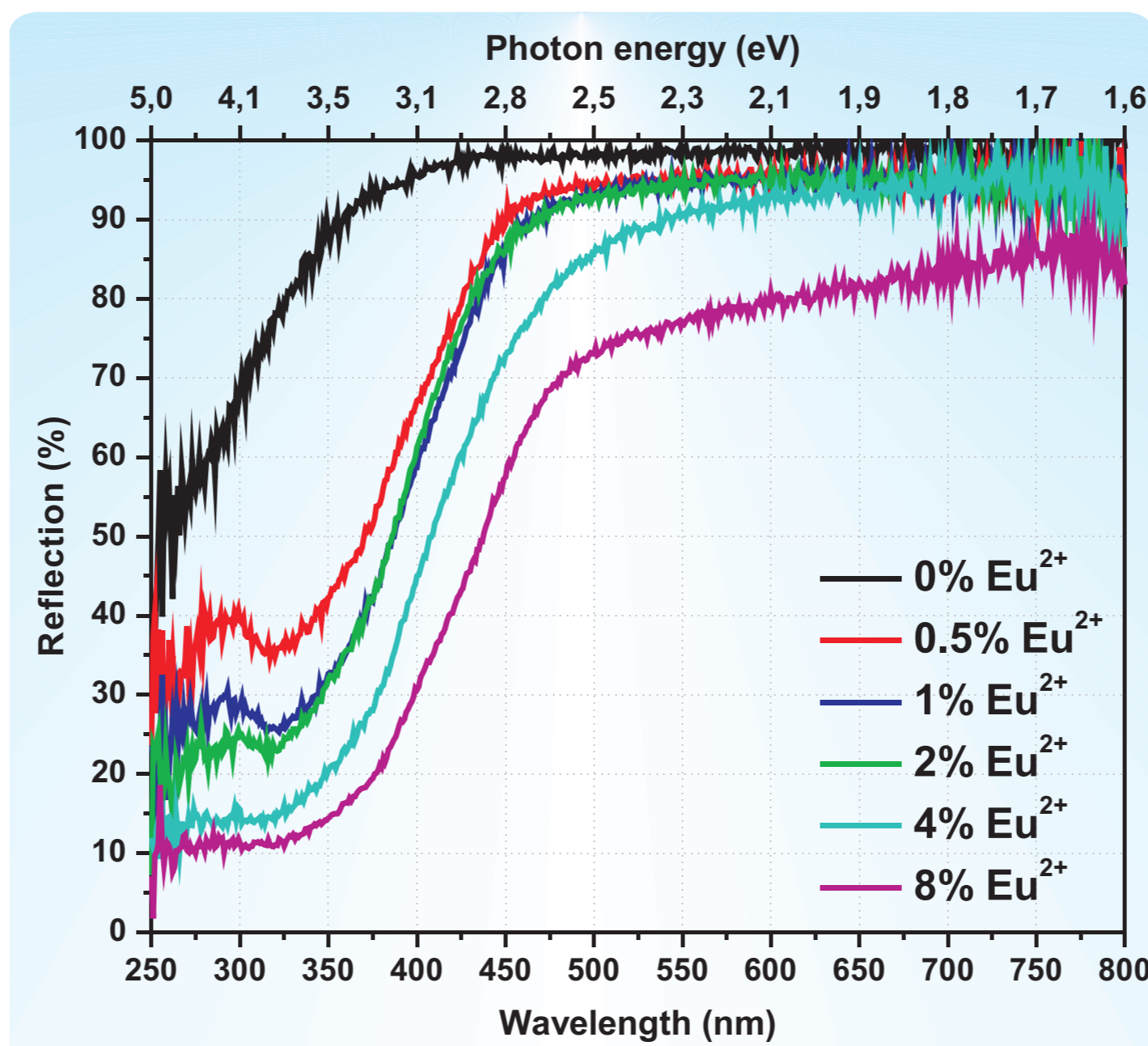


Fig. 2. Reflection spectra of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{Eu}^{2+}$ .

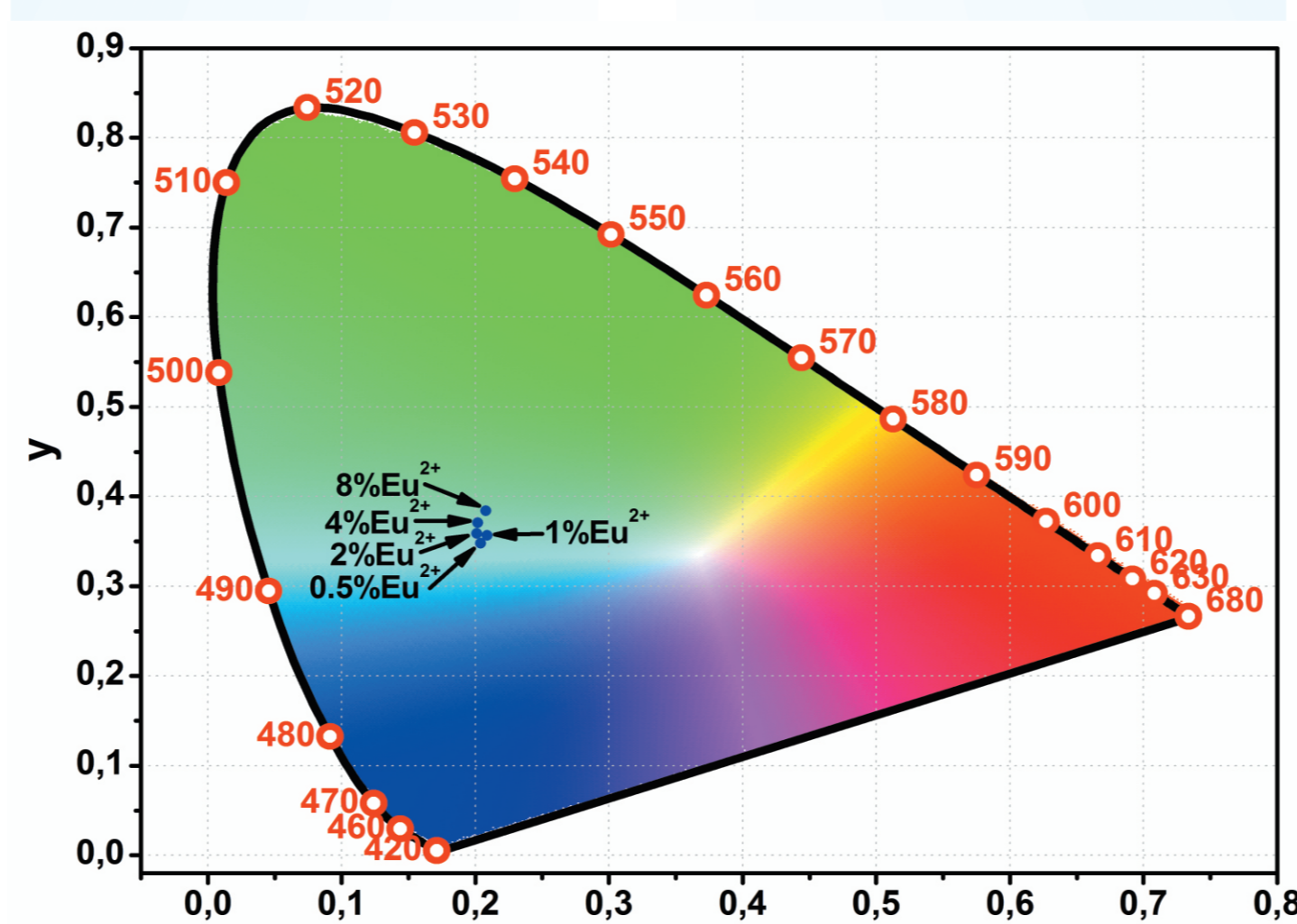


Fig. 5. CIE 1931 color points of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{Eu}^{2+}$ .

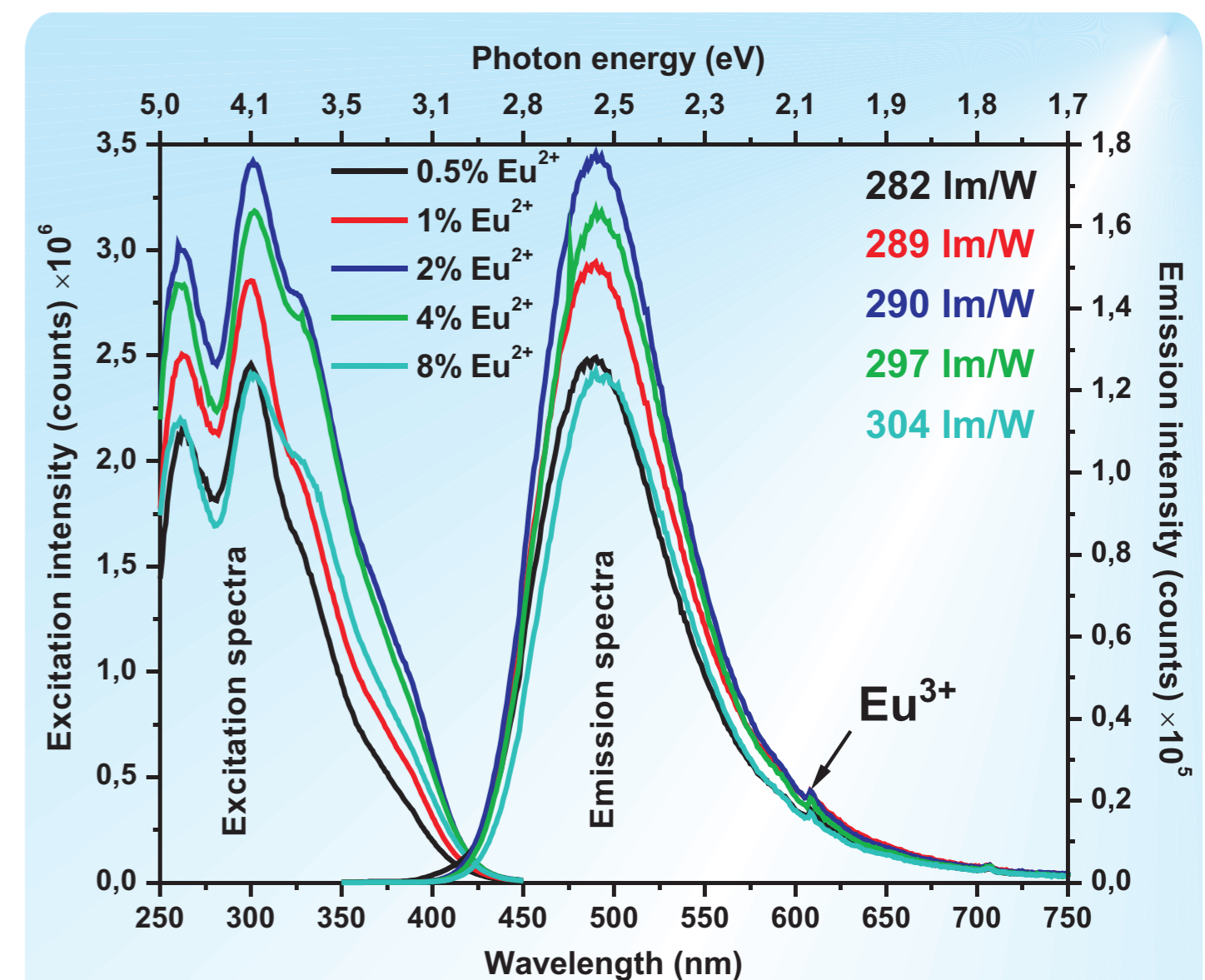


Fig. 3. Excitation and emission spectra of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{Eu}^{2+}$ .

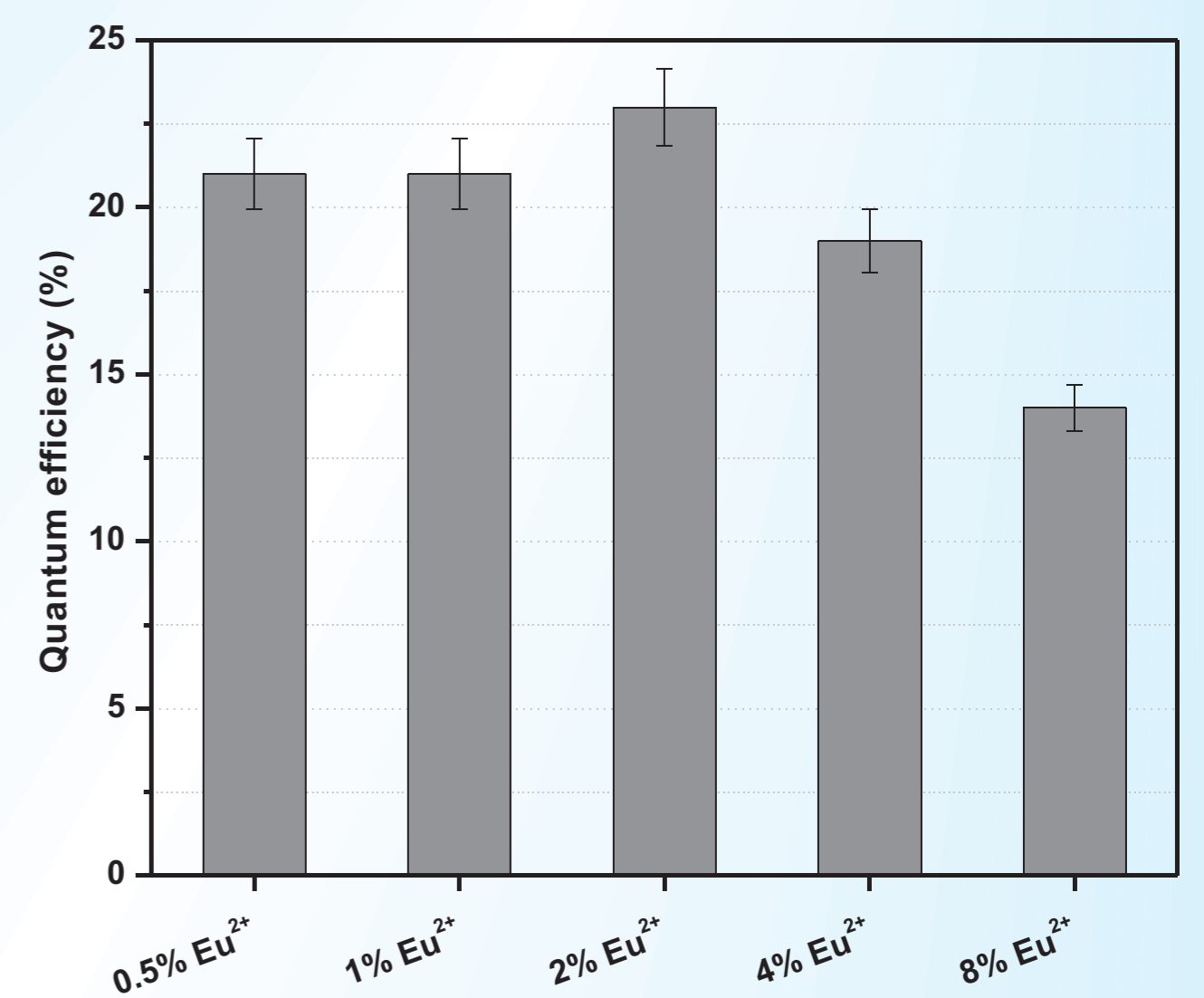


Fig. 6. Quantum efficiencies of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{Eu}^{2+}$ .

## Results and Discussion

XRD patterns of the synthesized samples are in good agreement with reference data regardless  $\text{Eu}^{2+}$  concentration (see Fig. 1). Fig. 2 shows reflection spectra of the target samples. As expected an absorption in the near UV region increases with higher  $\text{Eu}^{2+}$  concentration. The intensity of excitation and emission spectra reaches maximum (495 nm) if samples are doped with 2%  $\text{Eu}^{2+}$  (see Fig. 3). The temperature dependent emission spectra of  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}:\text{2\%Eu}^{2+}$  sample are depicted in Fig. 4. It turned out that emission intensity rapidly decreases with increased temperature. Calculations revealed that phosphor loses half of efficiency already at  $71^\circ\text{C}$ . CIE 1931 color points of synthesized phosphors are shown in Fig. 5. It is obvious that color points slightly shift towards green spectral region what is in line with emission spectra. Finally, Fig. 6 depicts quantum efficiencies of the synthesized phosphors as a function of  $\text{Eu}^{2+}$  concentration. Quantum efficiencies do not exceed 25% and is huge drawback of the phosphor for application in solid state lighting.

## Conclusions

In this work we demonstrated that  $\text{Eu}^{2+}$  doped  $\text{Ba}_2\text{Zr}_2\text{Si}_3\text{O}_{12}$  phosphor possess emission in cyan-green spectral region. Quantum efficiencies do not exceed 25%, which is likely caused by photoionisation of the excited state. This explanation is in line with the strong thermal quenching of this phosphor.