

6. High Pressure Discharge Lamps

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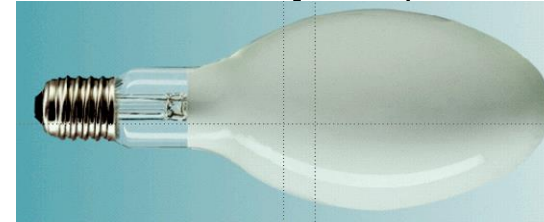
6.1 Overview of Low- and High-Pressure Discharge Lamps

HID = High Intensity Discharge

Hg low-pressure (TL)



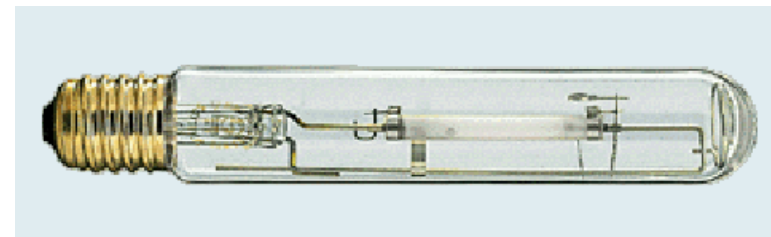
Hg high-pressure (HPMV = high pressure metal vapour)



Hg low-pressure (CFL, PL)



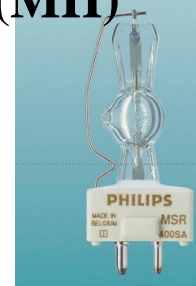
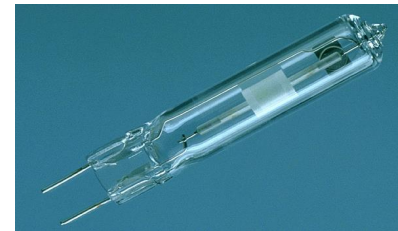
Na high-pressure (HPS = high pressure sod.)



Na low-pressure (SOX)

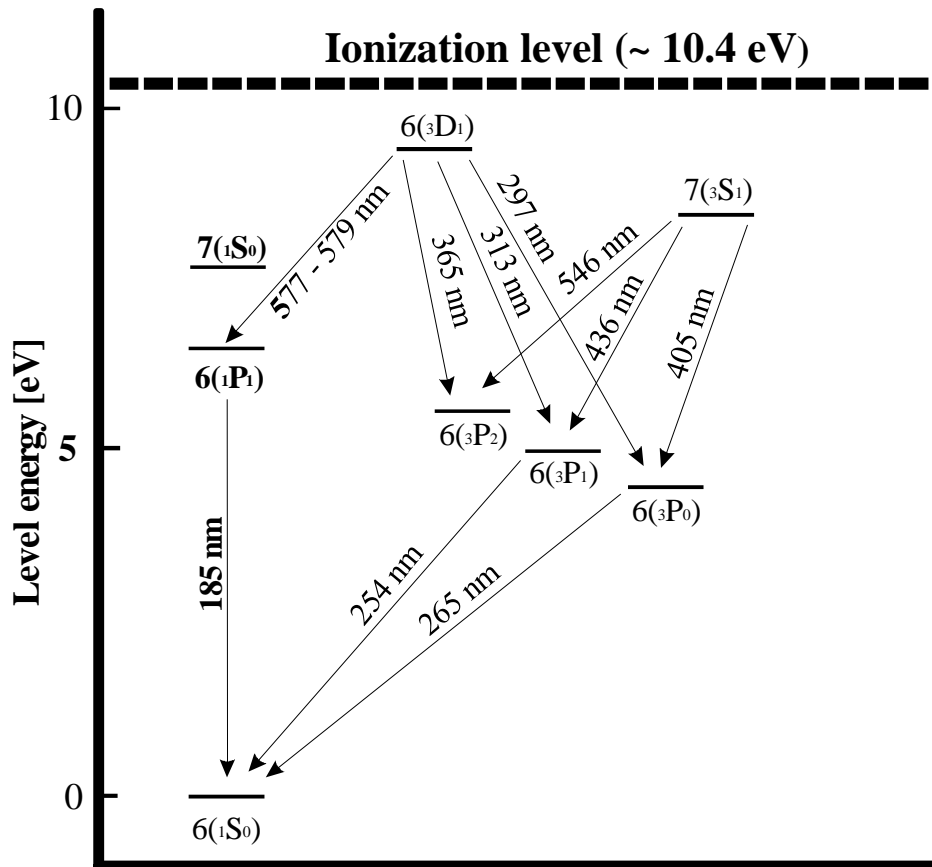


Metal-halide high-pressure (MH)

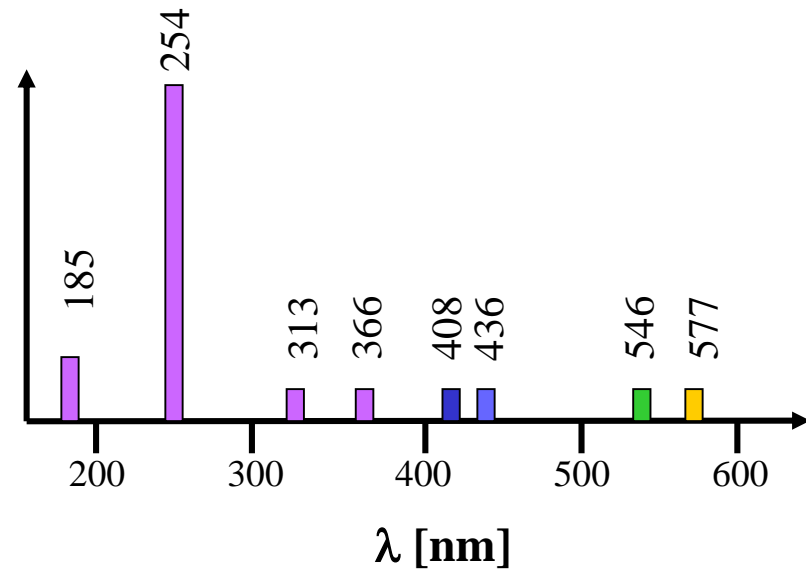


6.2 Spectrum of Hg Discharges

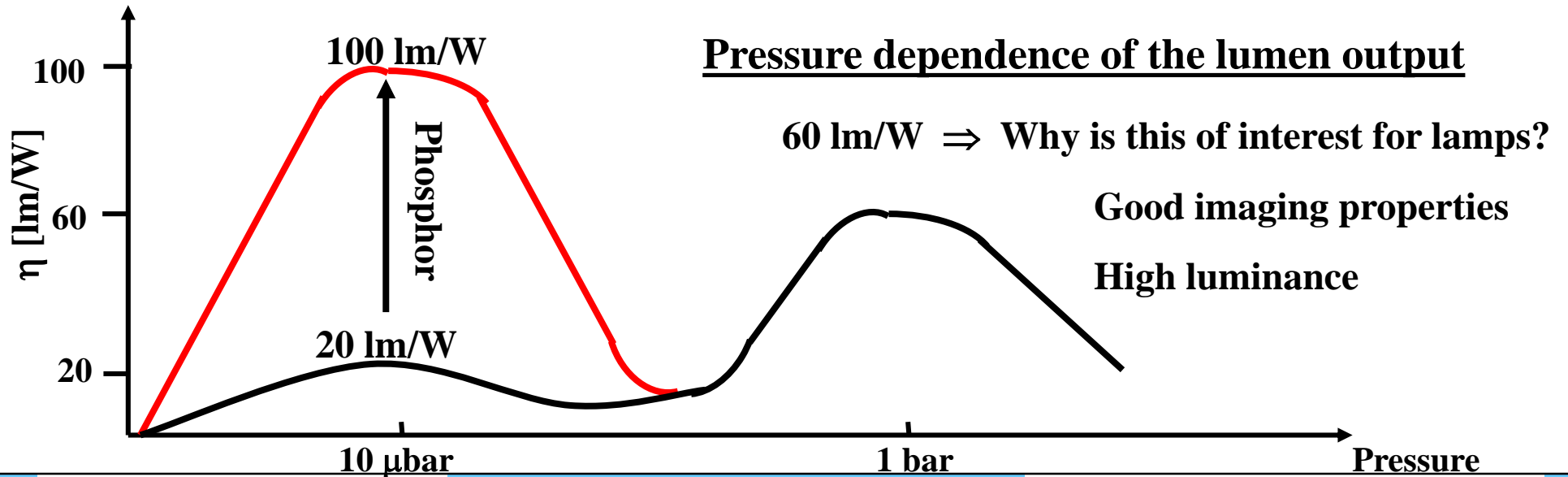
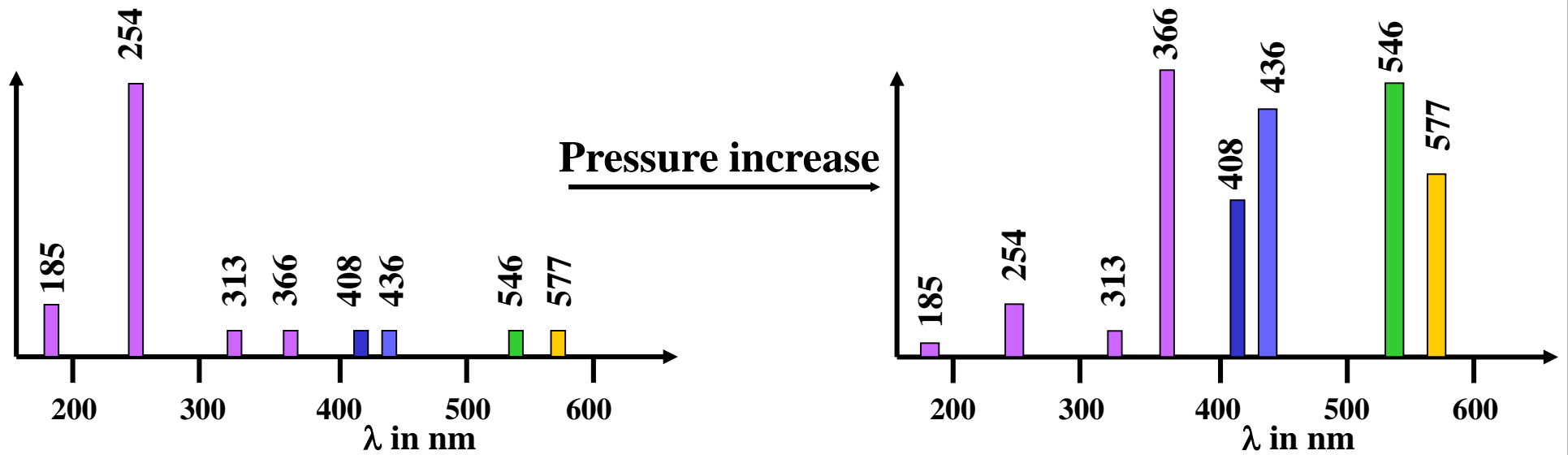
Energy level scheme of Hg



Schematic emission spectrum of a Hg discharge at a low pressure, i.e. in the mbar range



6.2 Spectrum of Hg Discharges



6.2 Spectrum of Hg Discharges

Measured spectra of a water-cooled capillary mercury discharge lamps

Broadening of emission lines due to

temperature increase

- Doppler broadening
- Rotational broadening

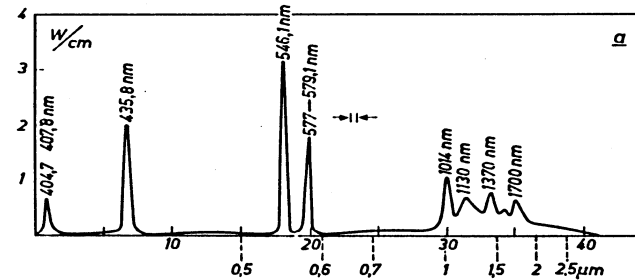
pressure increase

- Collisional broadening
- Reabsorption

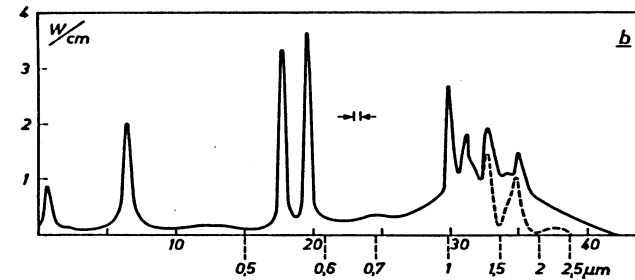
field strength increase

- Gravitational broadening
- Magnetical broadening

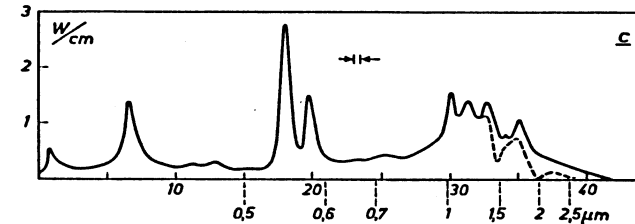
Source: W. Elenbaas, Quecksilberdampf-Hochdrucklampen (1966)



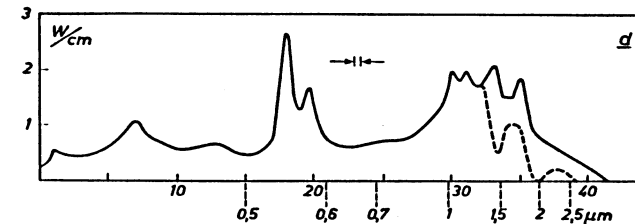
P = 25 atm.



P = 30 atm.

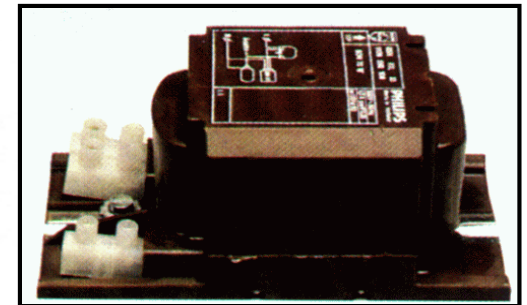
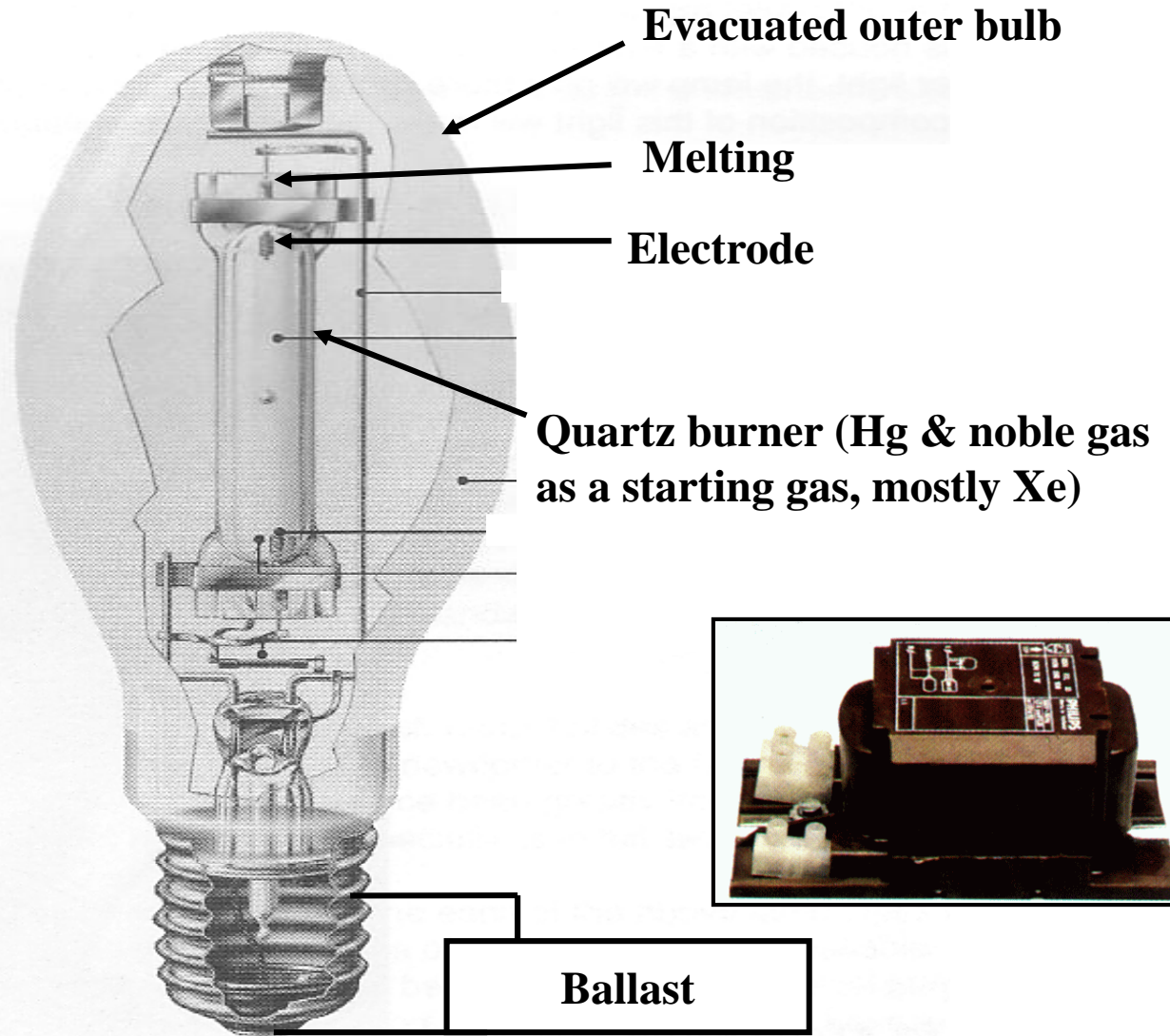


P = 100 atm.

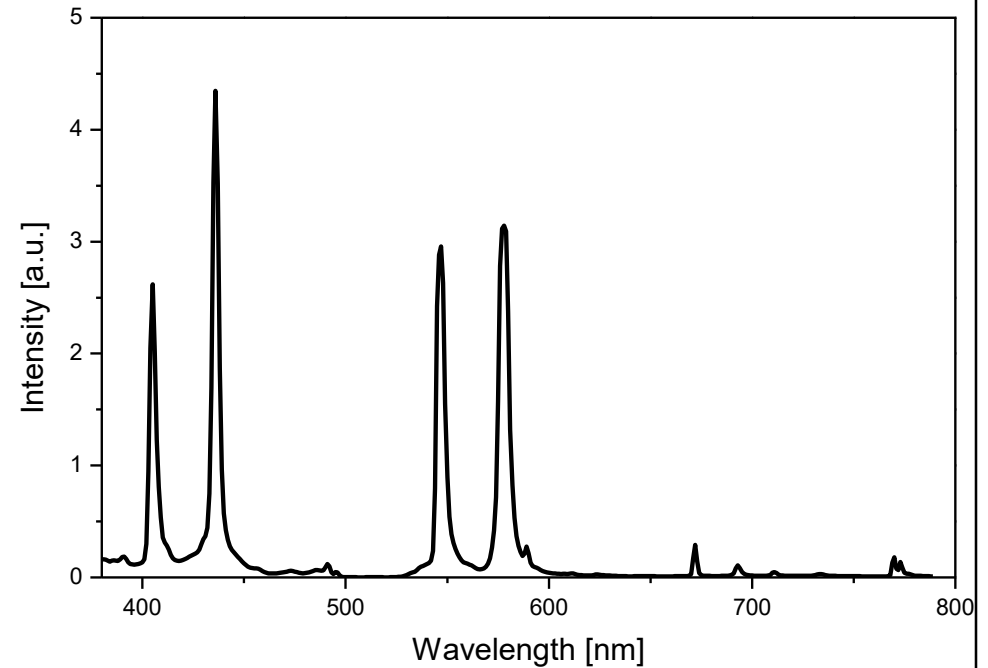
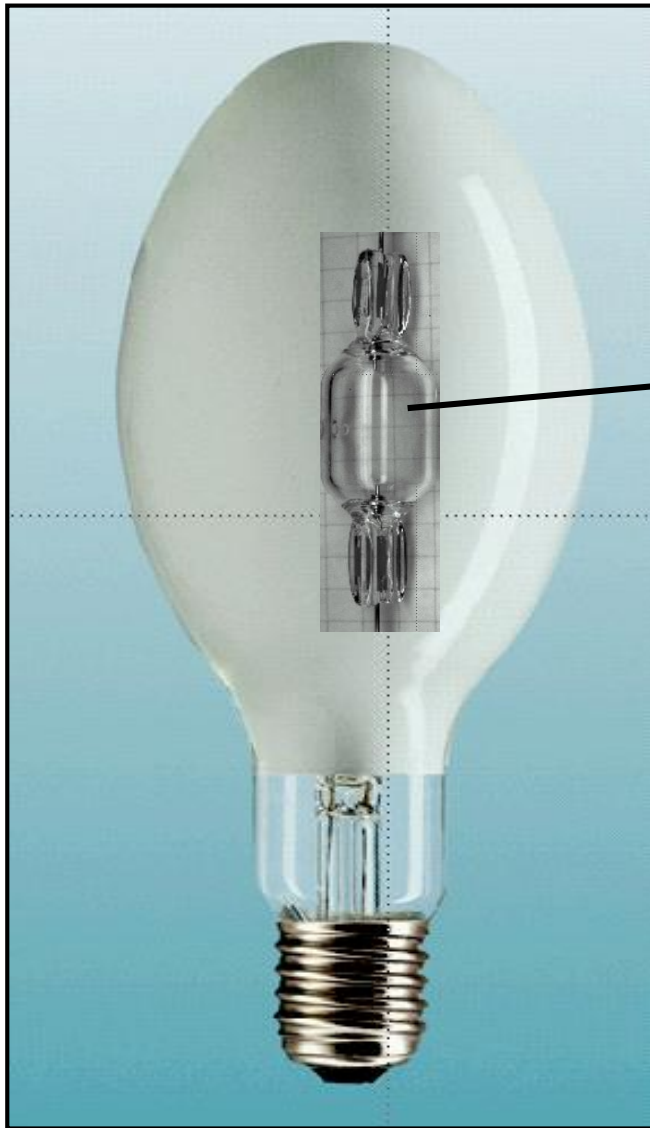


P = 150 atm.

6.3 The High-Pressure Mercury Lamp (HP)



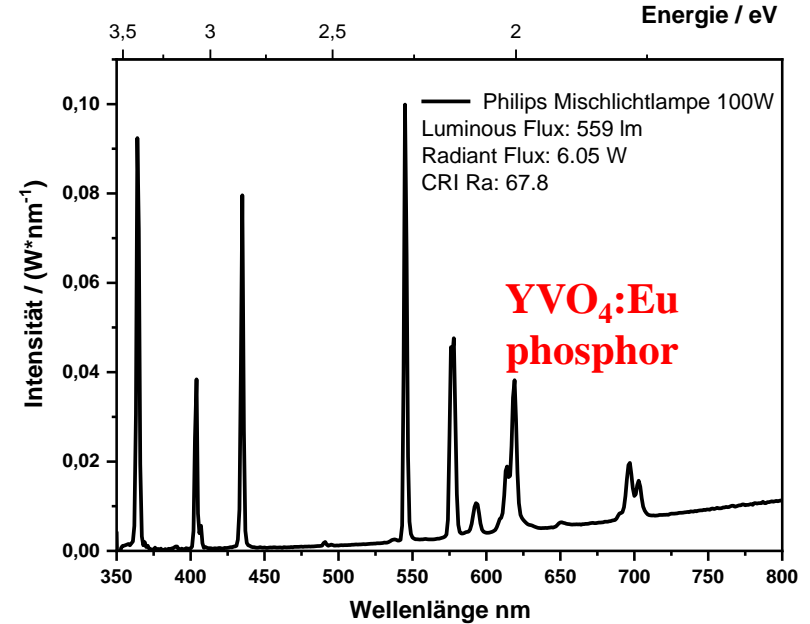
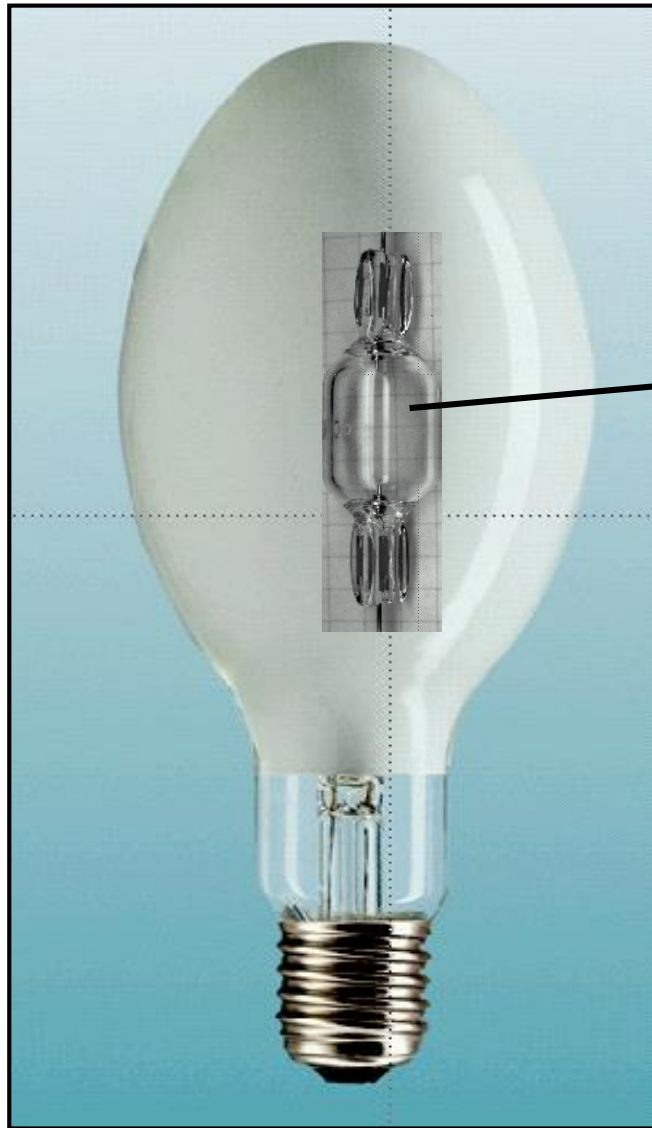
6.4 Phosphors for High-Pressure Mercury Lamps



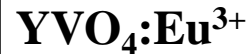
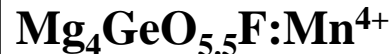
$\eta = 60 \text{ lm/W}$
 $R_a = 20$
Lifetime = 20.000 h

Blue-white light due to the lack of red radiation in the emission spectrum
Solution: Phosphor layer

6.4 Phosphors for High-Pressure Mercury Lamps



Suitable phosphors



$\eta = 60 \text{ lm/W}$

$R_a = 50-70$

Lifetime = 20.000 h

620 nm Broadband emission

660 nm Line emission

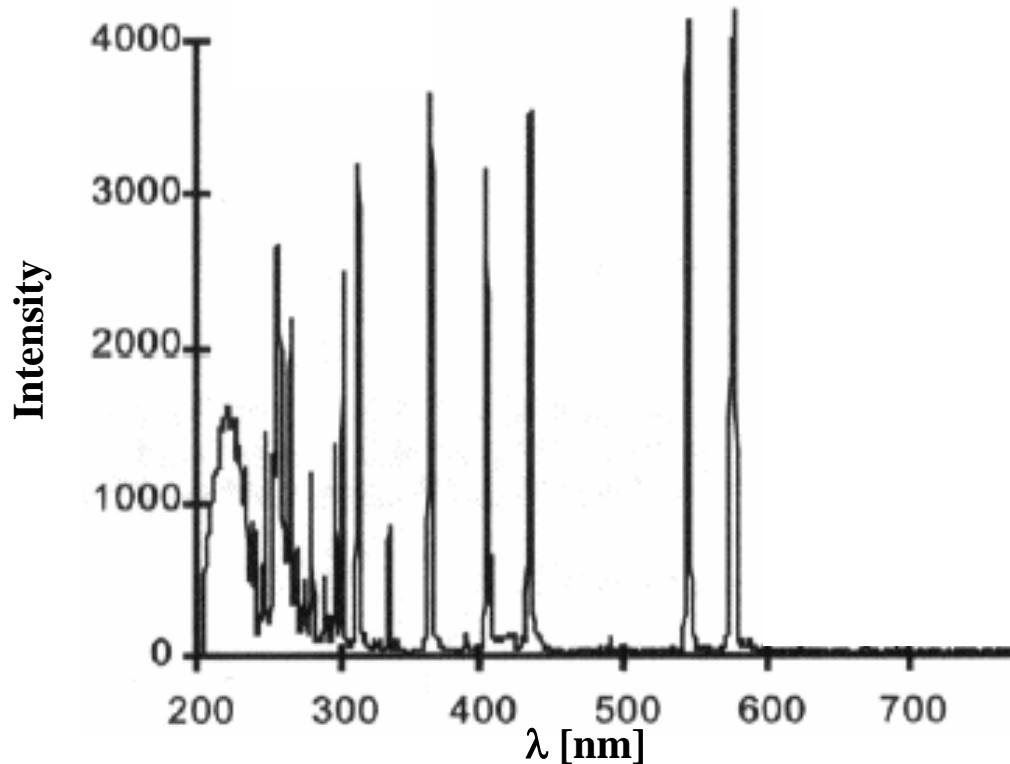
620 nm Line emission

620 nm Line emission

6.4 Phosphors for High-Pressure Mercury Lamps

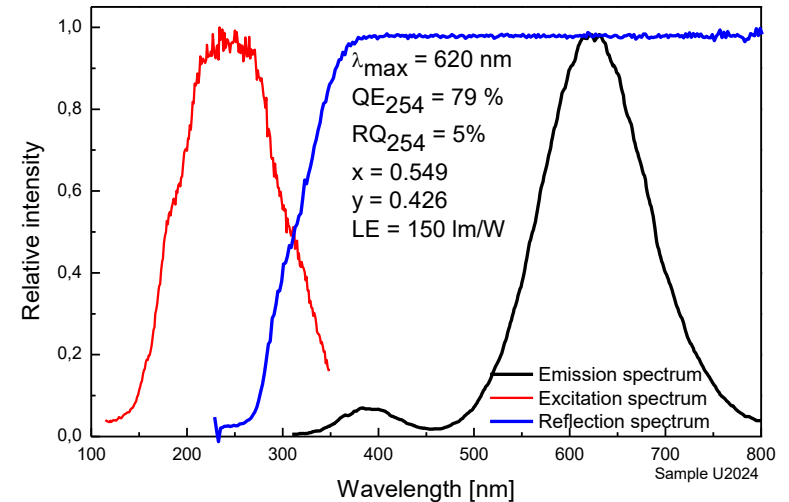
**Sn²⁺ or Mn⁴⁺ phosphors
as UV → Red converter**

Hg HP lamps emit UV radiation substantially

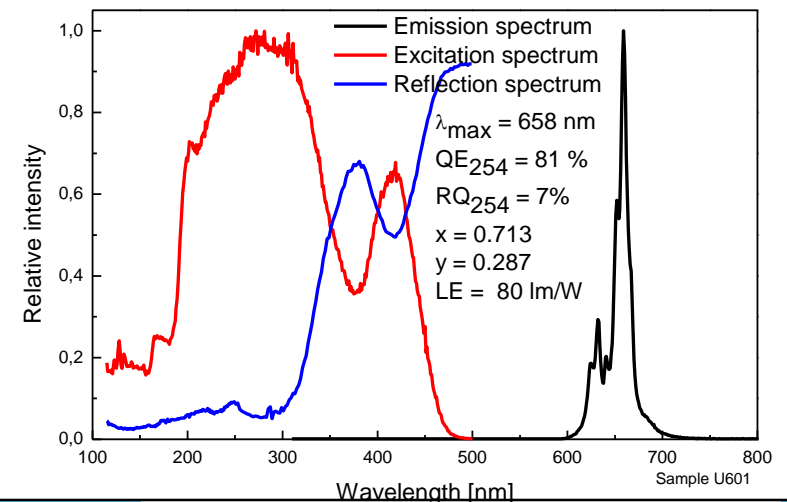


Problem: Low luminous efficacy of these phosphors

Luminescence spectra of (Sr,Mg)₃(PO₄)₂:Sn



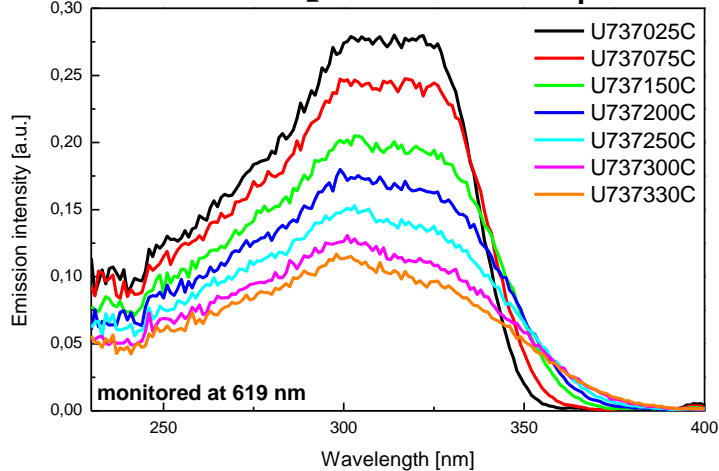
Luminescence spectra of Mg₄GeO_{5.5}F:Mn



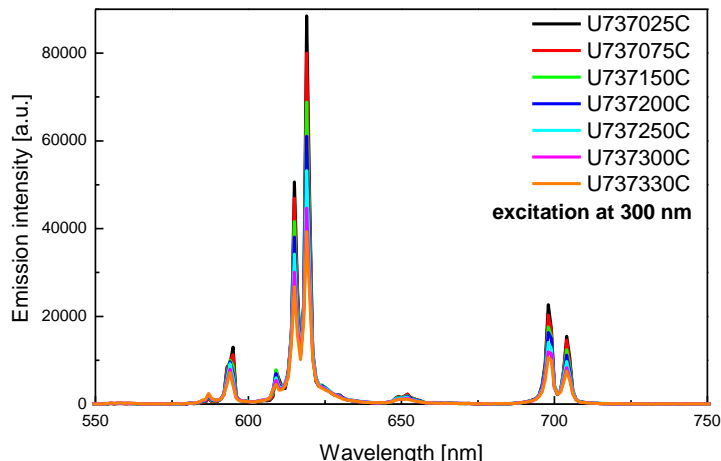
6.4 Phosphors for High-Pressure Mercury Lamps

YVO₄:Eu³⁺ phosphors - Thermal behavior

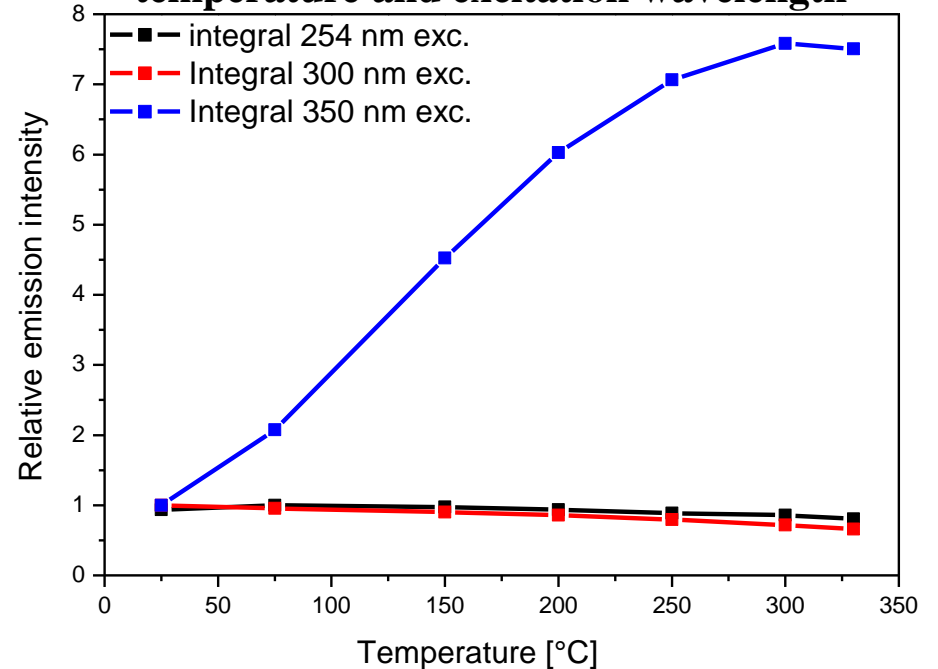
Excitation spectra of YVO₄:Eu



Emission spectra of YVO₄:Eu



Luminescence intensity as a function of temperature and excitation wavelength

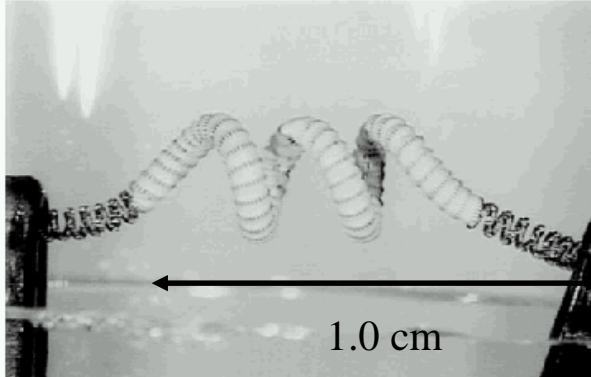


The luminous efficacy under UV-A excitation increases up to about 300 °C

Cause: Increase in spectral overlap with Hg high-pressure discharge emission spectrum

6.5 The Electrode

Hg low-pressure

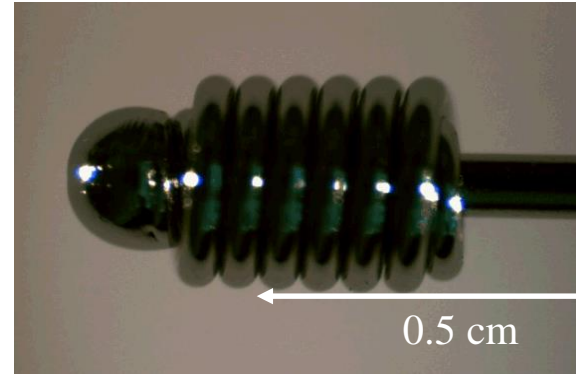


36 W
I = 0.36 A

Tungsten + emitter
BaO / SrO / CaO

T = 1350 K

Hg high-pressure



400 W
I = 4 A

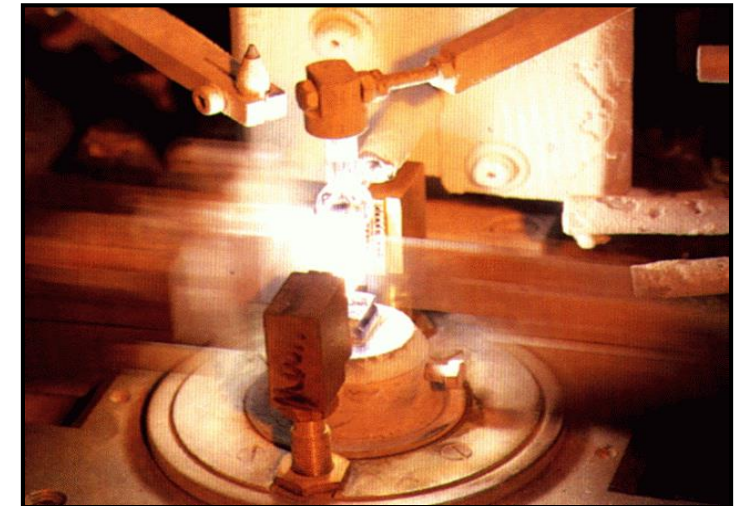
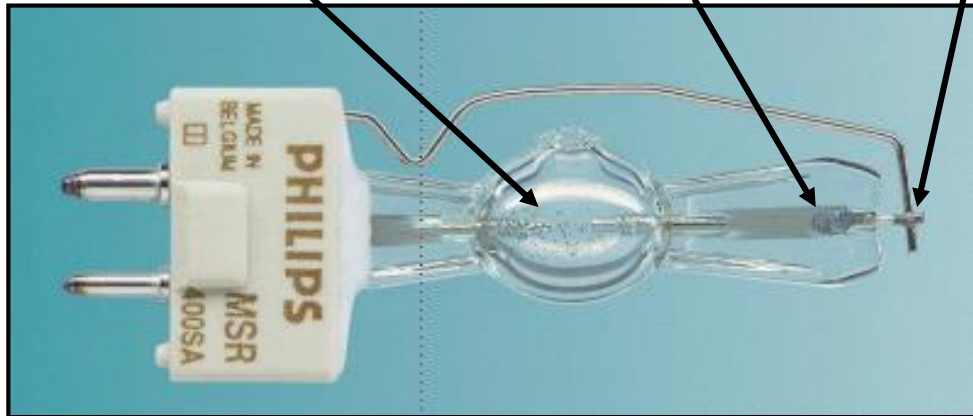
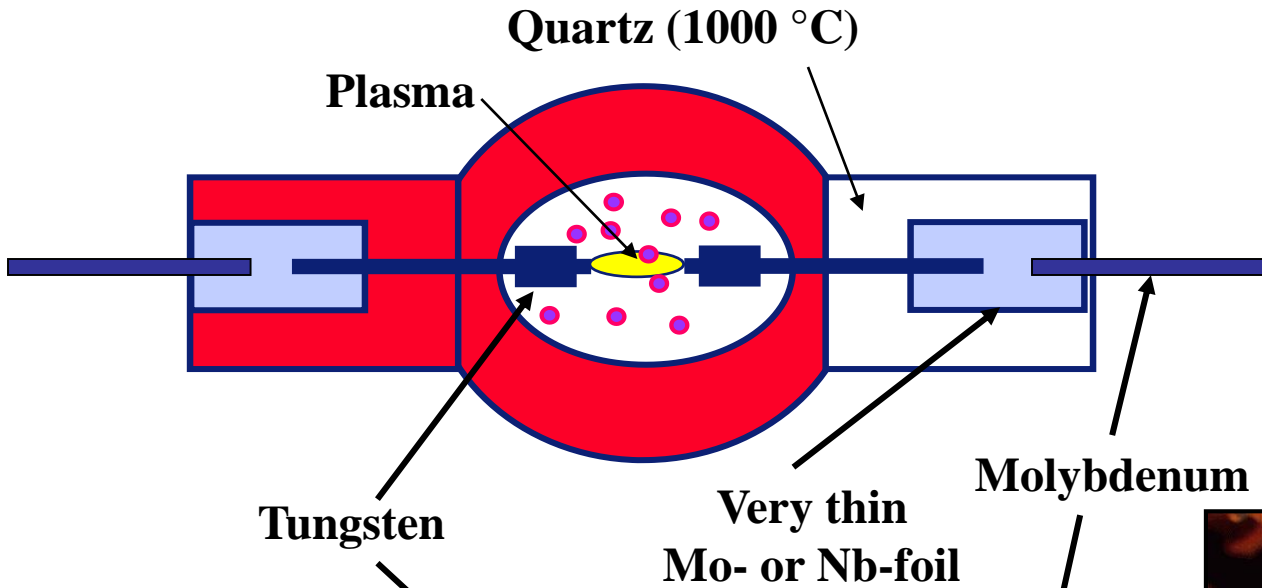
Tungsten + emitter
BaO / SrO / Y₂O₃ / ThO₂

T = 2000 - 3000 K

6.6 The Electrode Feedthrough

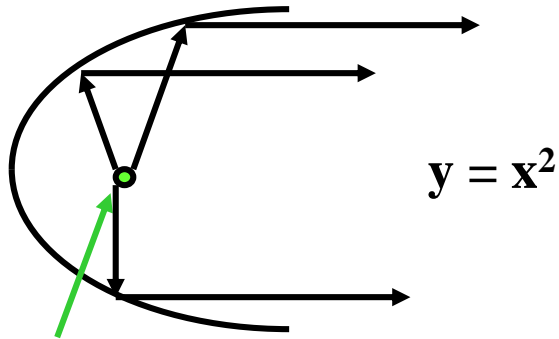
Problem: Different thermal expansion coefficients

SiO ₂	$\alpha = 0.5 \cdot 10^{-6} \text{ K}^{-1}$
W	$\alpha = 4.3 \cdot 10^{-6} \text{ K}^{-1}$
Mo	$\alpha = 2.8 \cdot 10^{-6} \text{ K}^{-1}$
Nb	$\alpha = 7.3 \cdot 10^{-6} \text{ K}^{-1}$



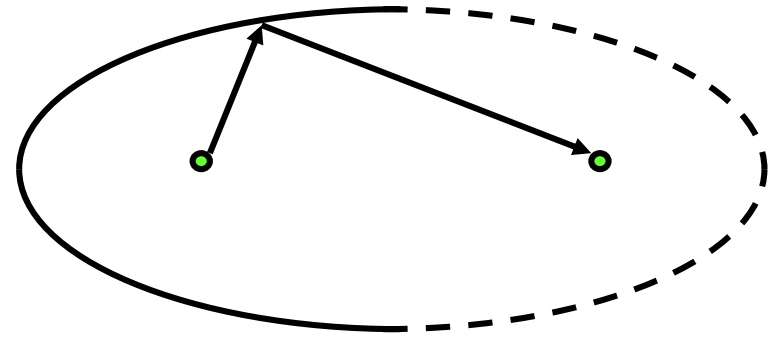
6.7 Types of Reflectors

Parabolic reflectors



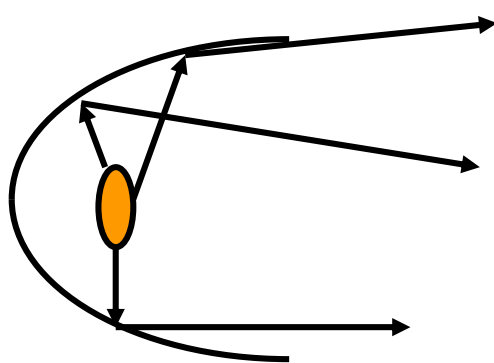
Focal point (light source)

Elliptical reflectors

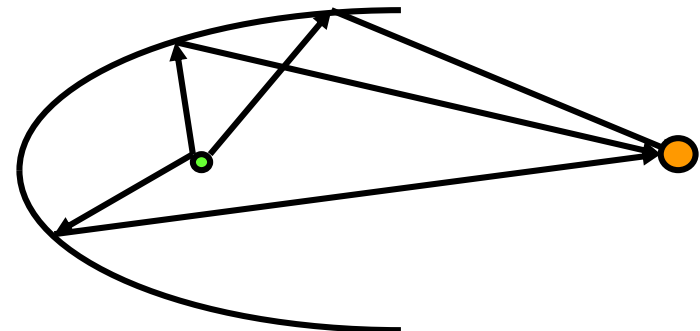


An ellipse has two focal points, one is occupied by the light source

Only possible if the light source is point like

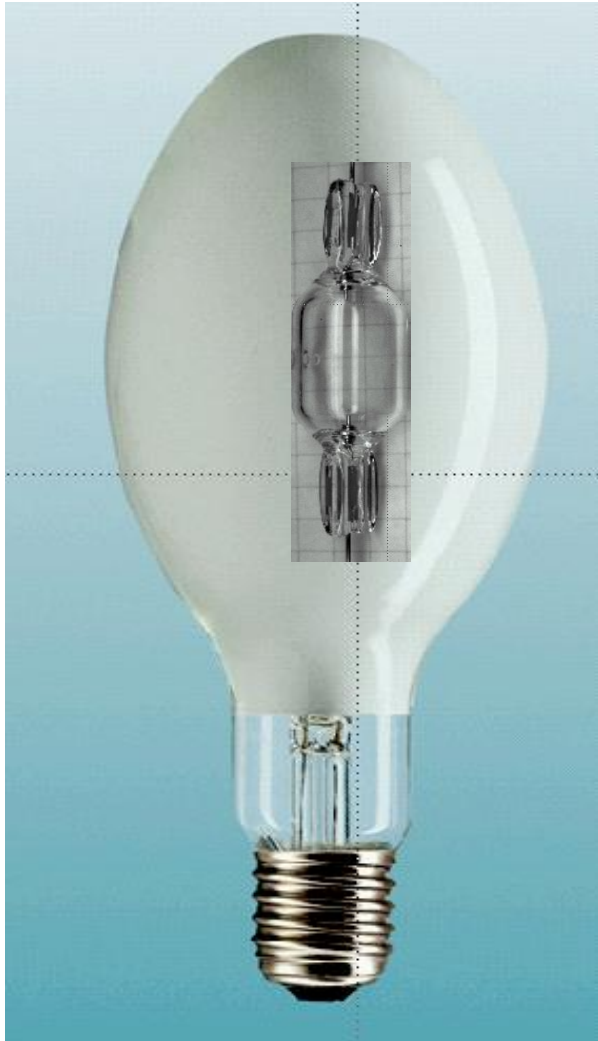


HID lamps

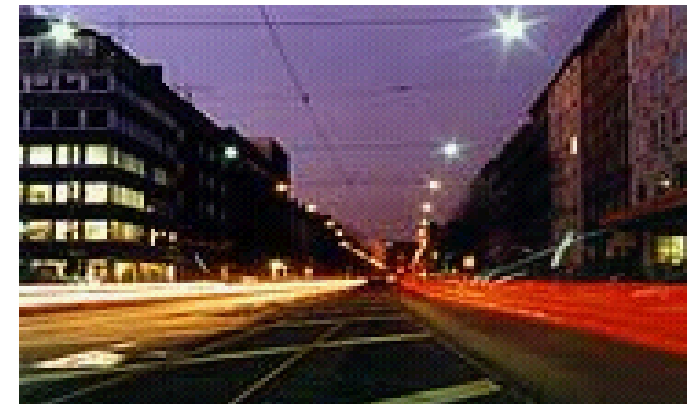


6.8 Application of HP-Lamps

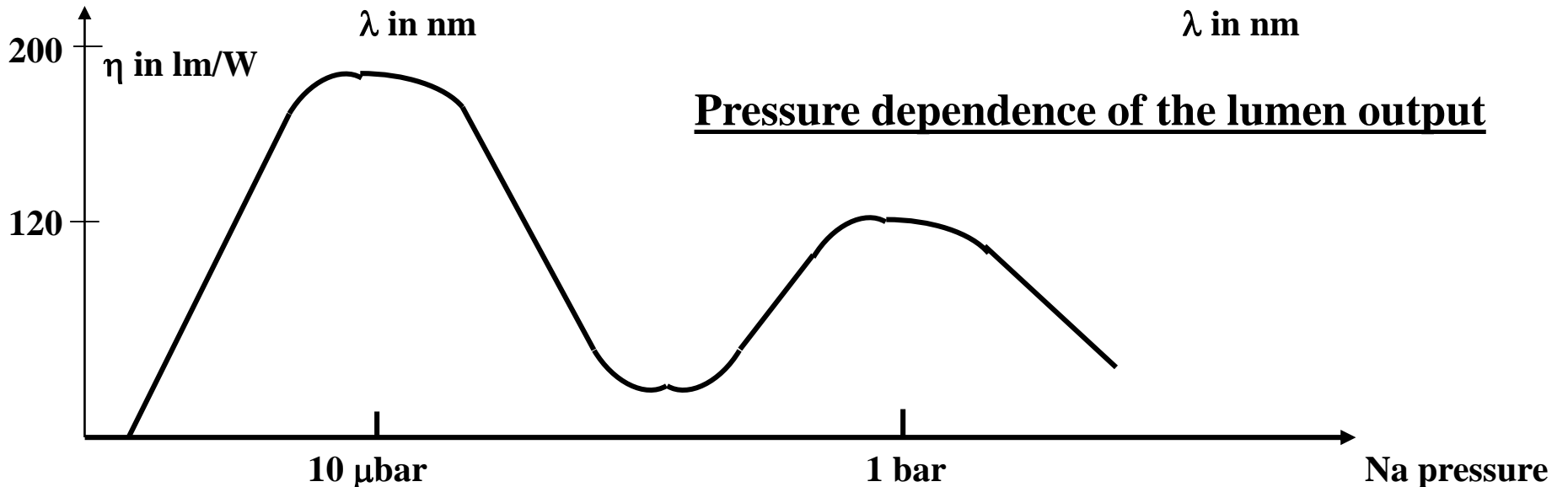
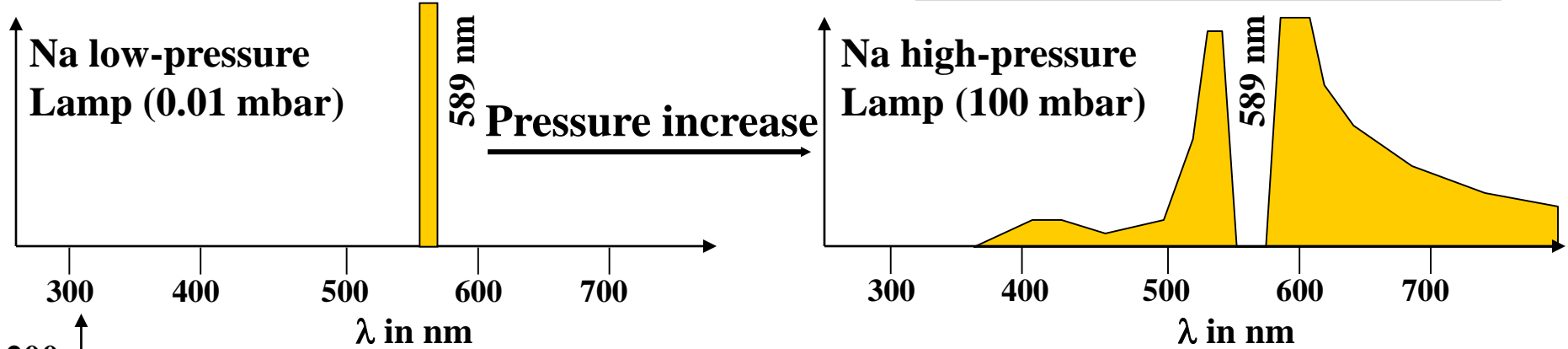
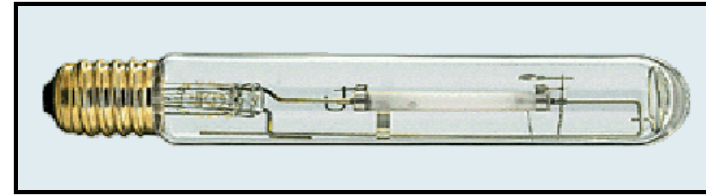
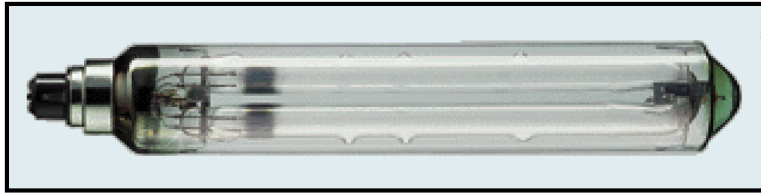
In street lighting (outdoor lighting)



$\eta = 60 \text{ lm/W}$
 $R_a = 50$
Lifetime = 20.000 h
P = 100 W - 2000 W

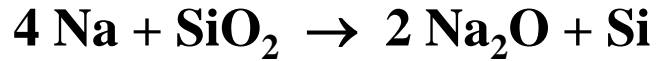


6.9 The High-Pressure Sodium Lamp (HPS)



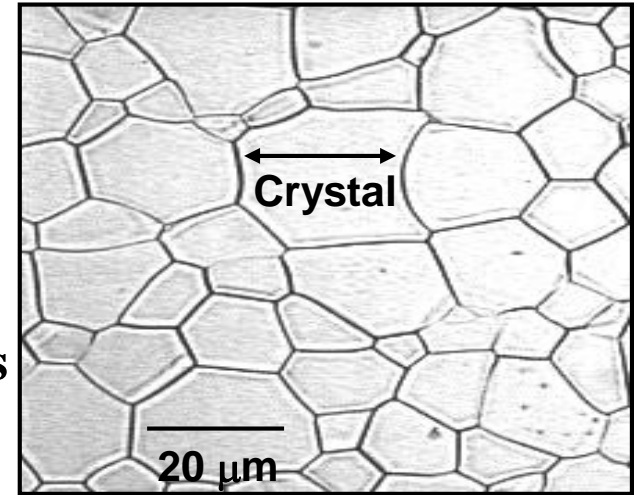
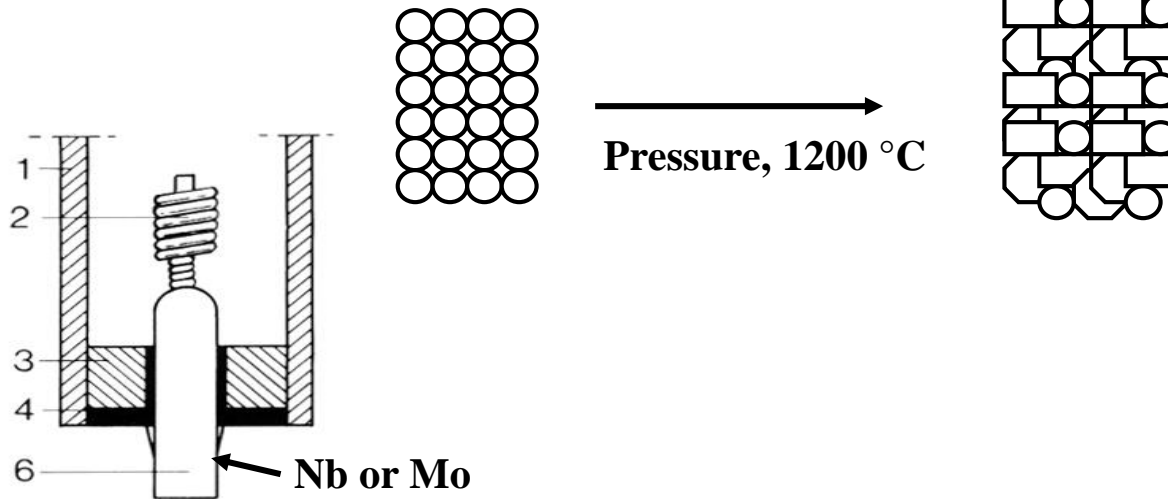
6.9 The High-Pressure Sodium Lamp (HPS)

Problem: Na reacts at high temperatures with the quartz glass wall



Solution: Transparent, high temperature resistant material, which does not react with Na

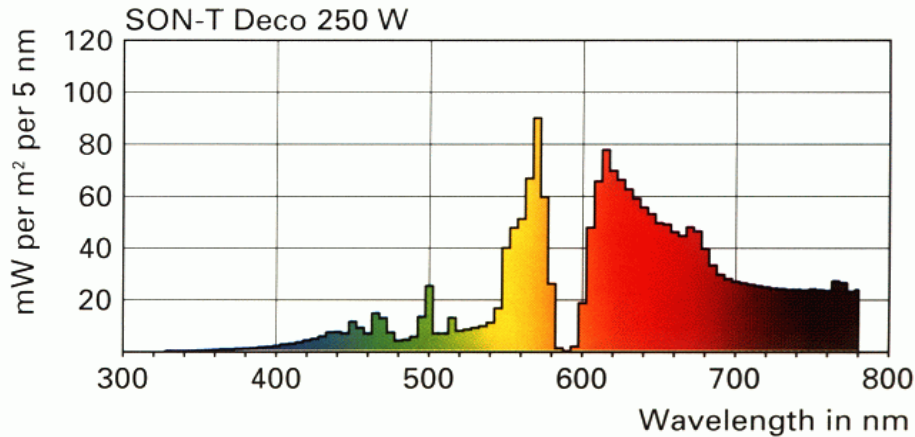
Al₂O₃-ceramics (corundum): MgO, CaO, B₂O₃-Additives (DSA = Densely Sintered Alumina)



Polycrystalline structure



6.9 The High-Pressure Sodium Lamp (HPS)



Widening of the Na-line and self-absorption leads to a spectral hole in the emission spectrum at around 589.3 nm

$p_{\text{Na}} = 150$ mbar (saturated)

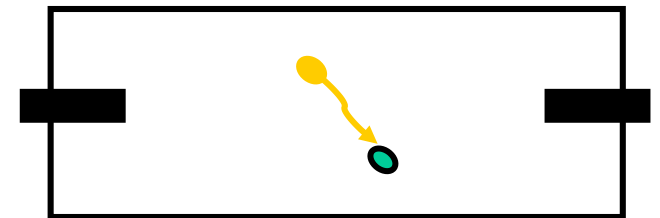
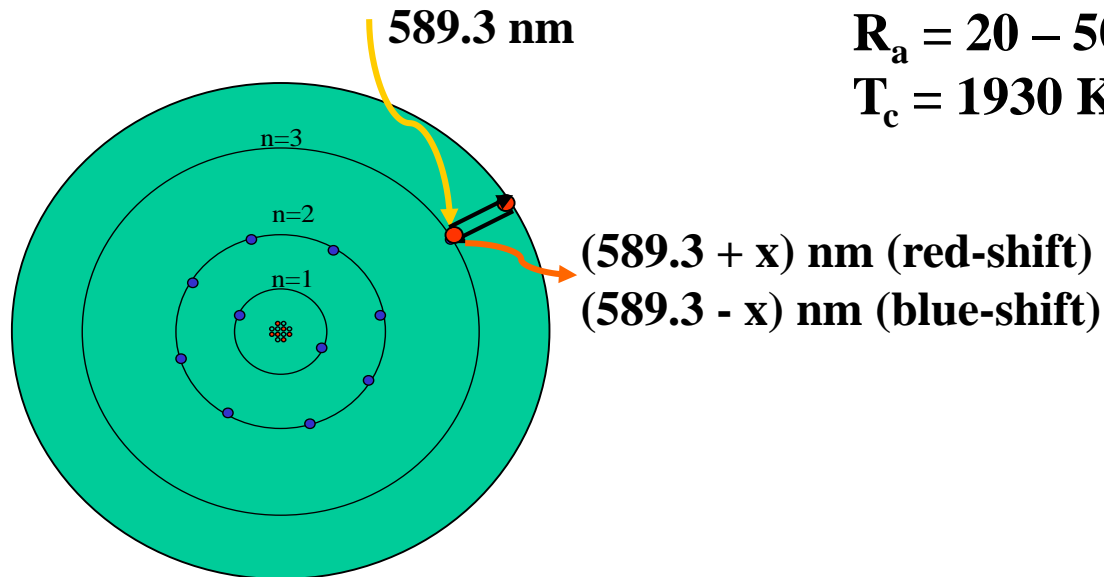
$p_{\text{Hg}} = 1000$ mbar (buffer gas)

$p_{\text{Xe}} = 100$ mbar (start gas)

$\eta = 90 - 120$ lm/W

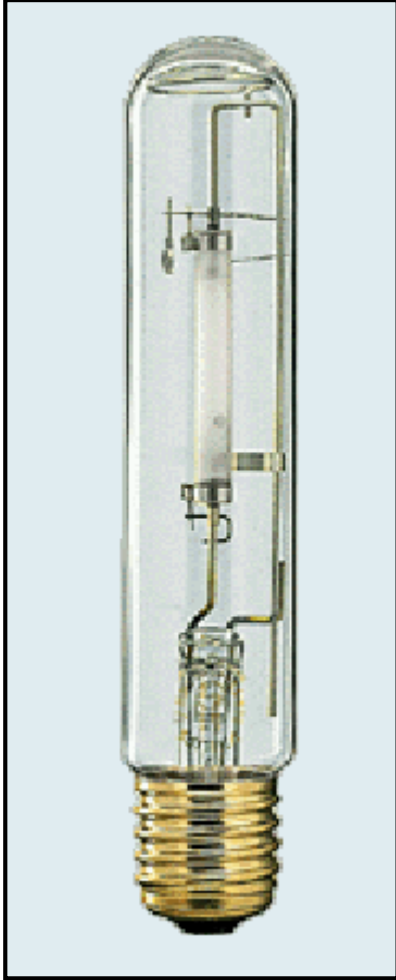
$R_a = 20 - 50$ (pressure dependent)

$T_c = 1930$ K



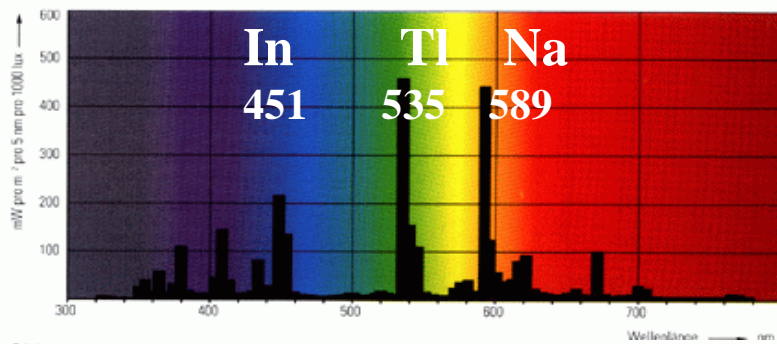
6.10 Application of HPS Lamps

Architectural and street lighting

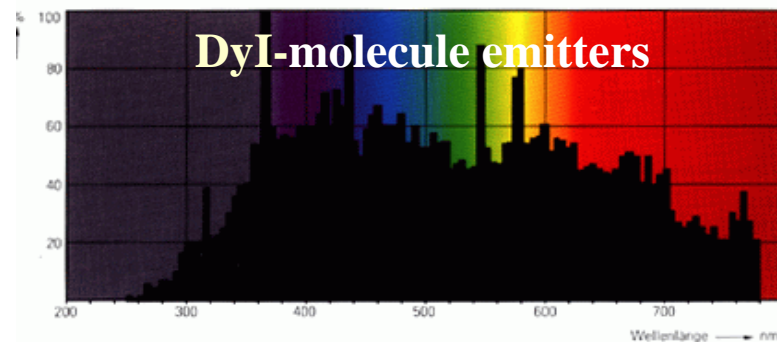
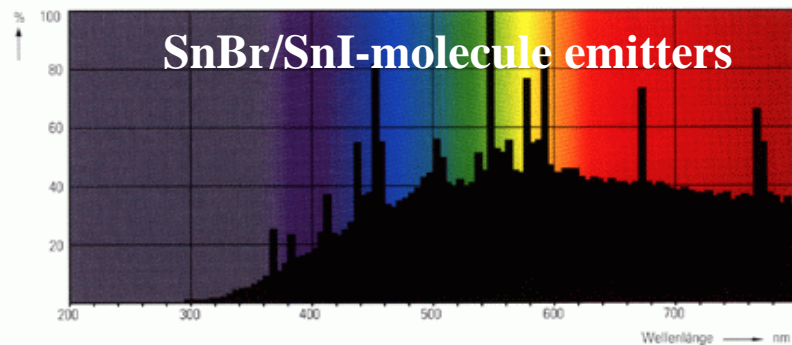


6.11 Metal-Halide High-Pressure Lamps

HPI 400 W BU



SN



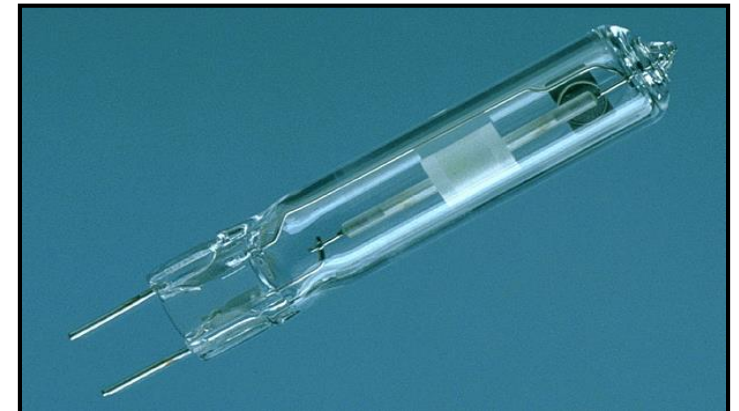
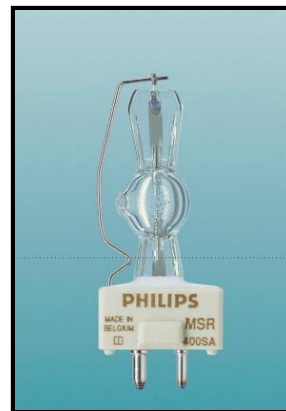
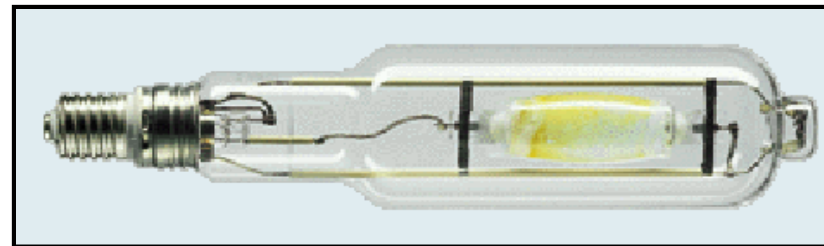
Filling: NaI - TlI - InI

SnBr₂ - SnI₂

NaI - DyI₃ (Studio-Stage-TV)

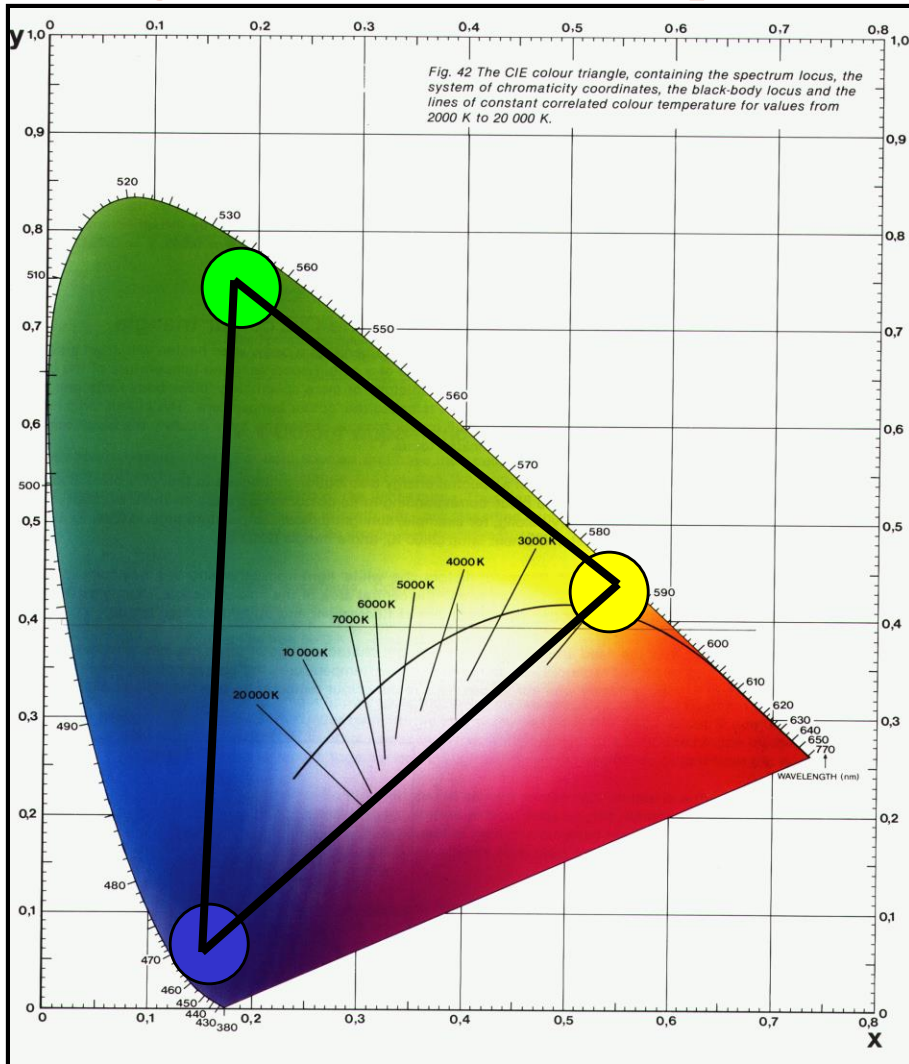
NaI - ScI₃ (automobile headlight)

Goal: High η & color rendering



6.11 Metal-Halide High-Pressure Lamps

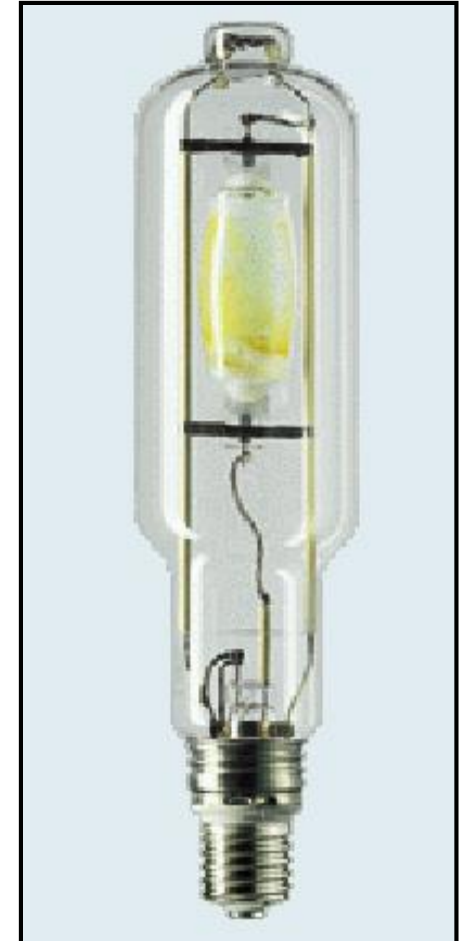
HPI (High Pressure Iodide) lamps



451 nm
(In)

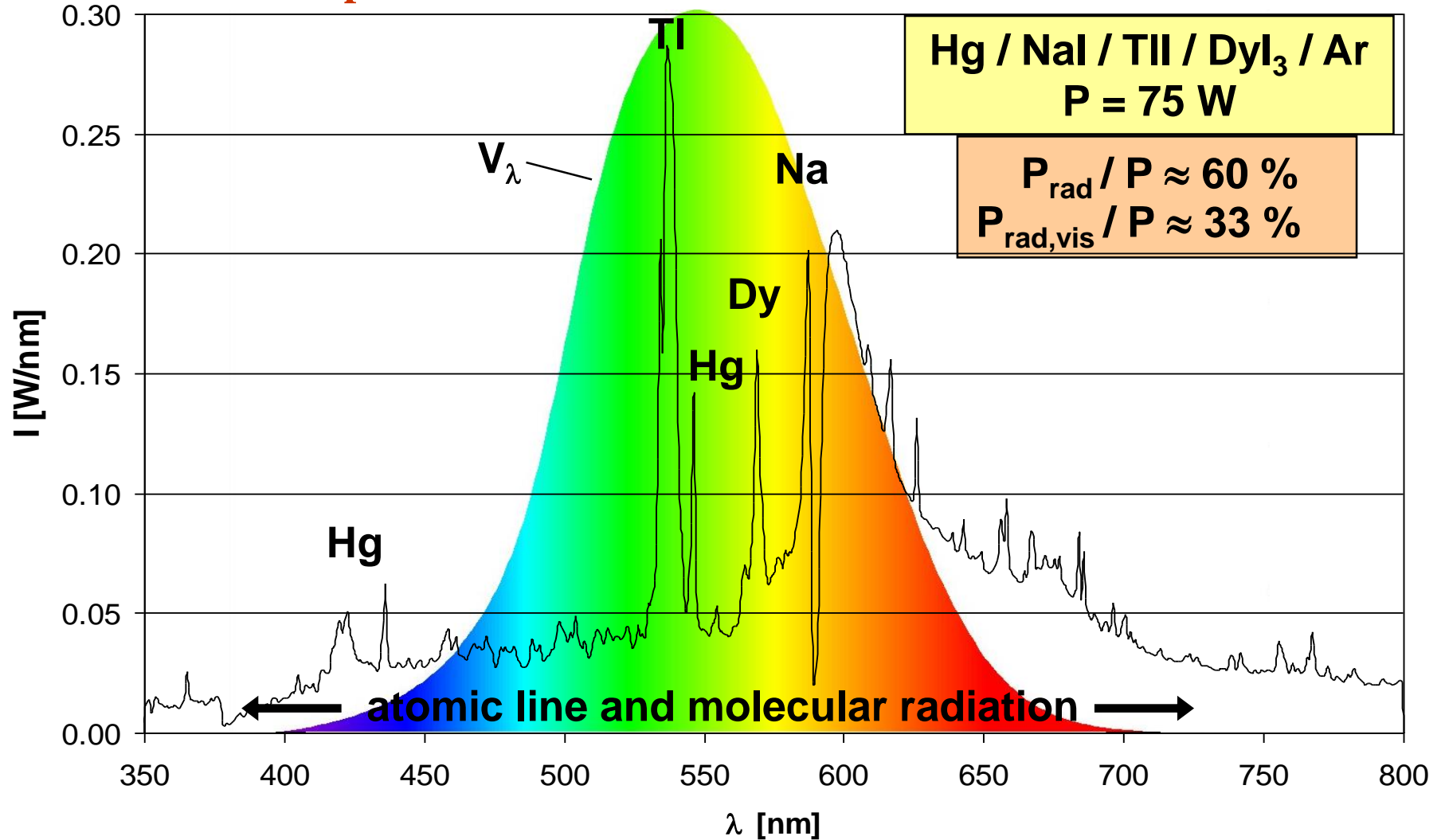
535 nm
(Tl)

589 nm
(Na)



6.11 Metal-Halide High-Pressure Lamps

Spectrum of a MH lamp



6.11 Metal-Halide High-Pressure Lamps

Filling of metal halide lamps

Lamp starting (starting gas)

Noble gases: Ar or Xe (xenon lamps) → Penning effect

Radioactive substances: ^{85}Kr , ^{147}Pm , ThO_2

Operating voltage

- Hg
- Trend towards the substitution of Hg (environmental aspect) → Zn

Light emission

- Hg
- Metal halides MeX_n (Me = Na, In, Tl, Sc, Sn, Dy, ...)

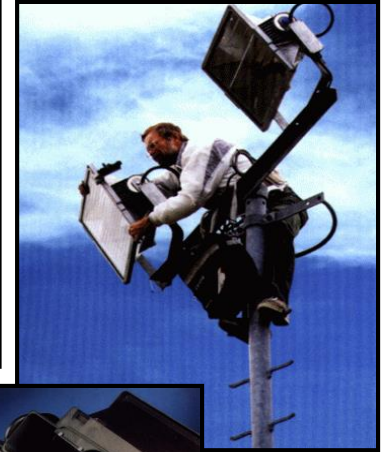
6.12 Photometric Data in Comparison

	Improvement	η (lm/W)	R_a	Color temperature T_c [K]
High Pressure Hg		60	20	6000
	+ phosphor	60	50	3800
High Pressure Na		60 - 130	20	2000
	Xe-pressure \uparrow	80 - 150	20	2000
	Na-pressure \uparrow	60 - 90	60	2200
Metal Halide	HPI (NaI-TlI-InI)	70 - 80	70	3800 - 4200
	SnBr ₂ -SnI ₂	70	90	
	NaI-DyI ₃	75 - 80	90	3800 - 5600
	NaI-ScI ₃	80 - 90	75	3600 - 4200

6.13 Applications of MH Lamps

**HPI
(NaI-TII-InI)**

**Street lighting
Architectural lighting
Sports field lighting**

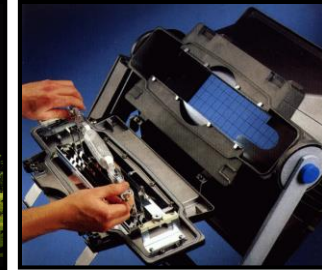


Tin

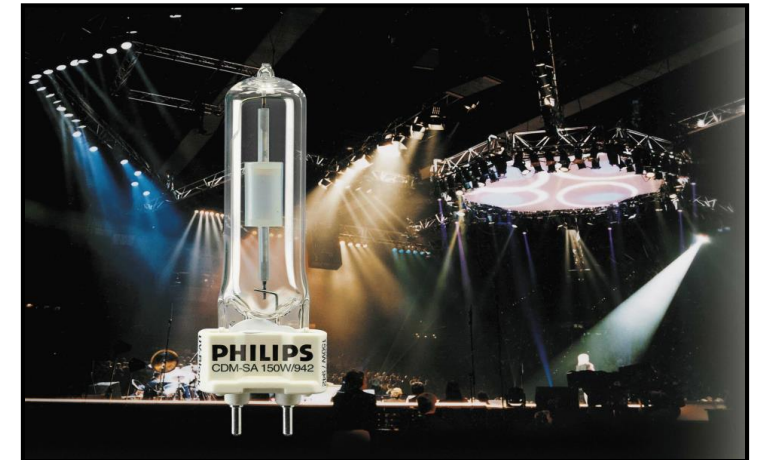
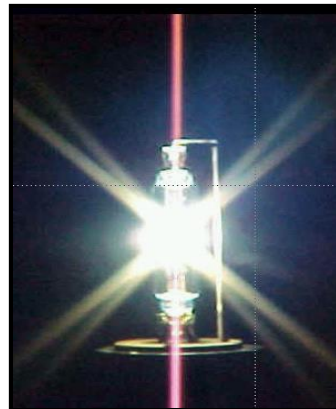
**Older type of lamp
is replaced by MH**

**NaI-DyI₃
NaI-ScI₃**

**Sports field lighting
Shop lighting
Studio-stage-TV (SSTV)
Automotive headlights**

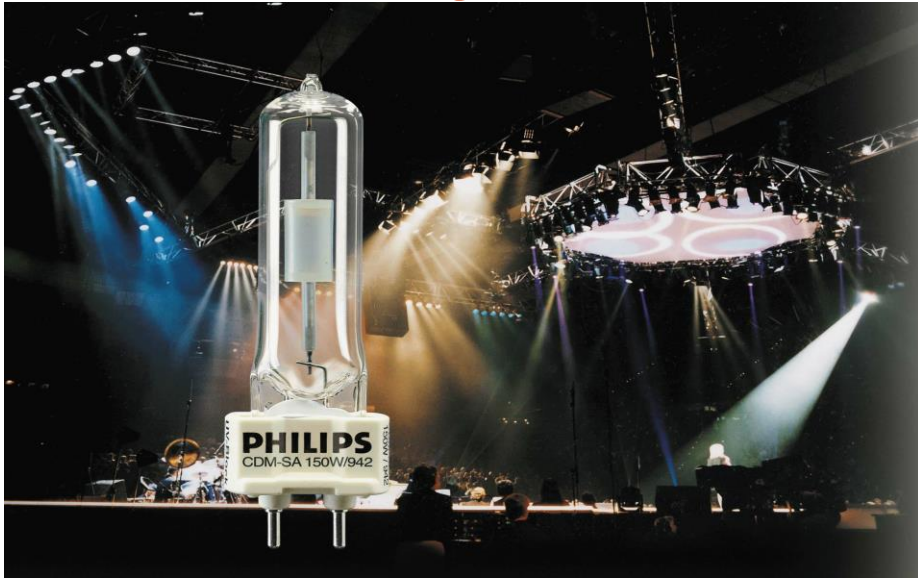


NaI-ScI₃ + Hg + Xe (blue)

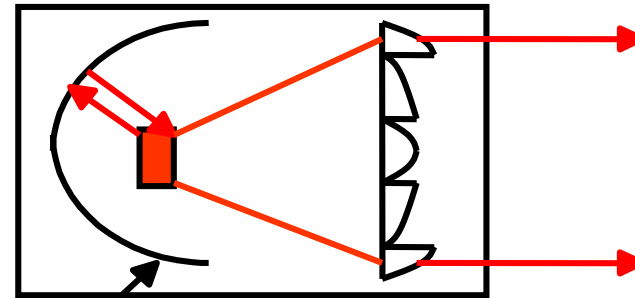


6.13 Applications of MH Lamps

SSTV market = Stage, Studio, and Television

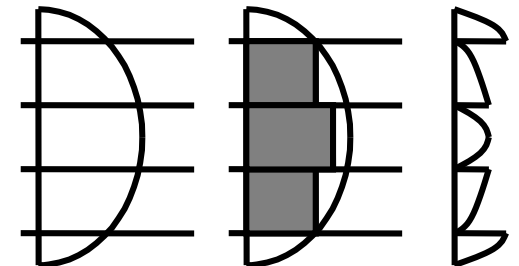


Reflector

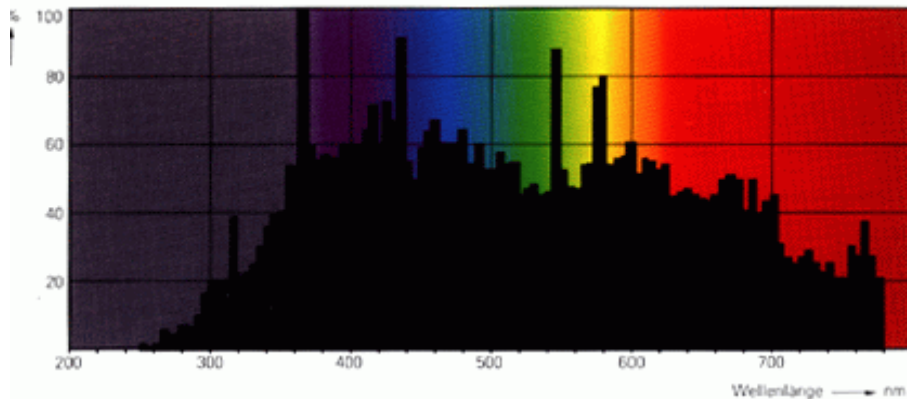


Spherical mirror

Fresnel-lens



MSR 400



6.13 Applications of MH Lamps

In the „beamer“

Warum Projektion ?

- Vorteile:
 - sehr große Bilder
 - kleines Volumen und Gewicht



**Rückwärts-
Projektion**



**Professionelle
Präsentationen**

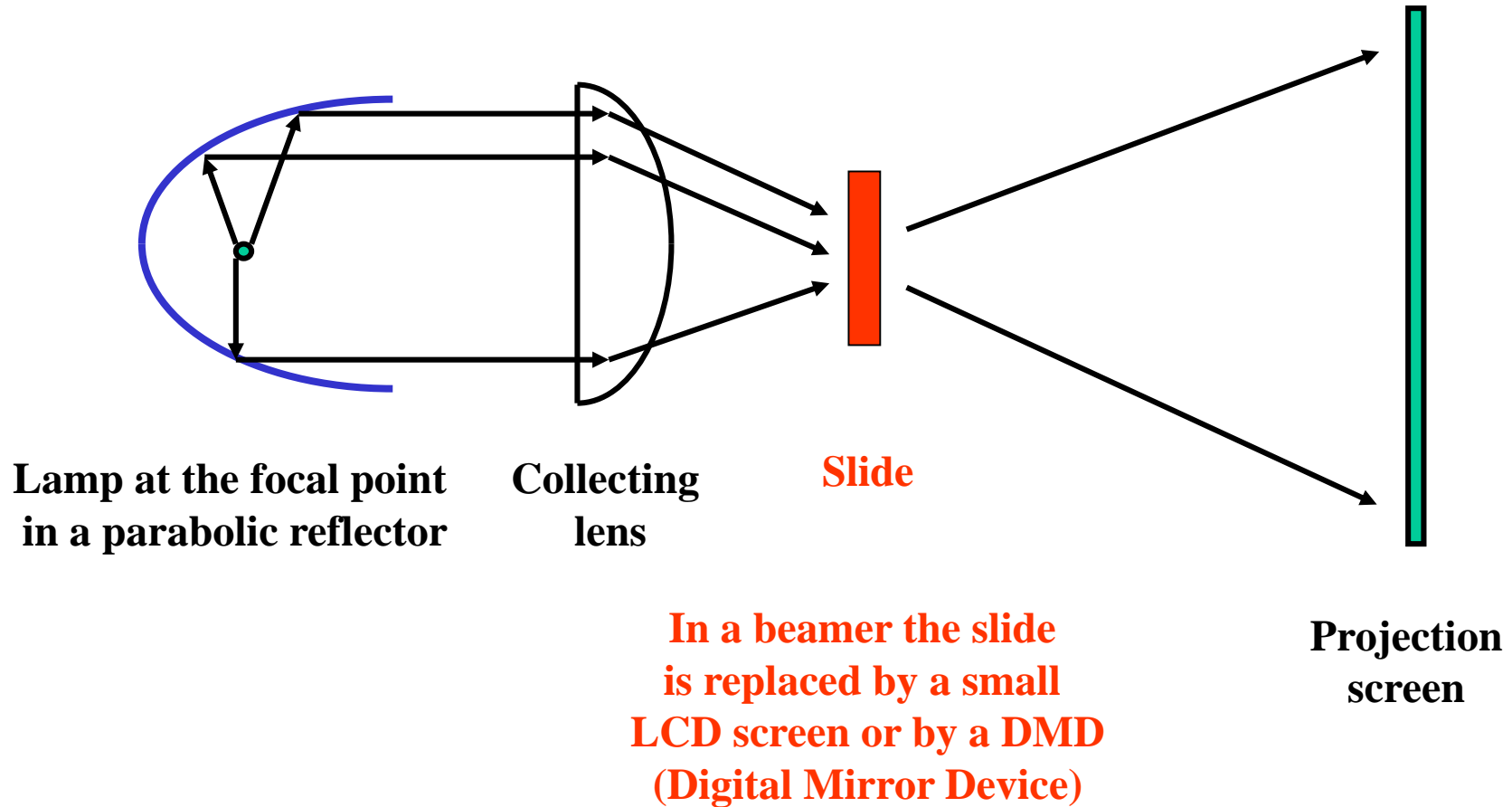


Heimkino

6.13 Applications of MH Lamps

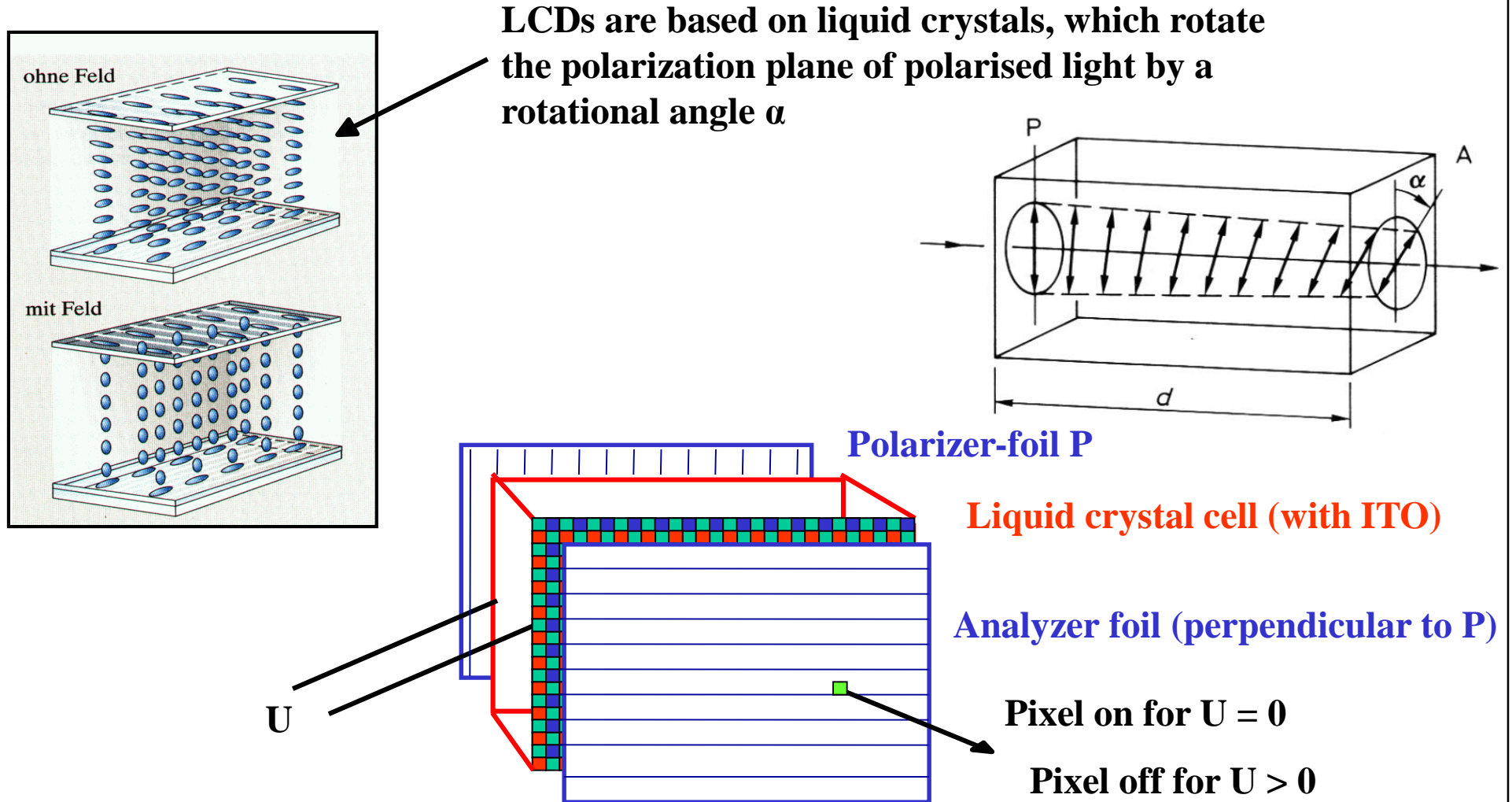
Construction of a beamer

A projector is actually a slide projector (diascope)!



6.13 Applications of MH Lamps

Operating principle of a LCD (Liquid Crystal Display)



6.14 UHP-Lamps

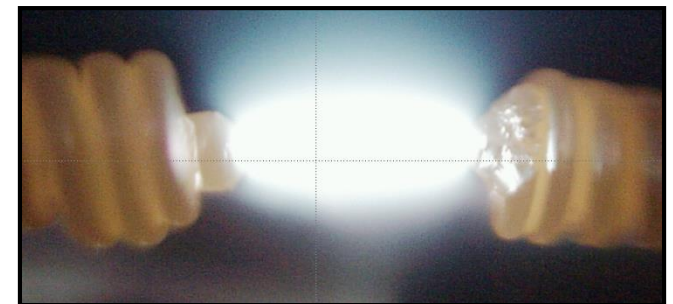
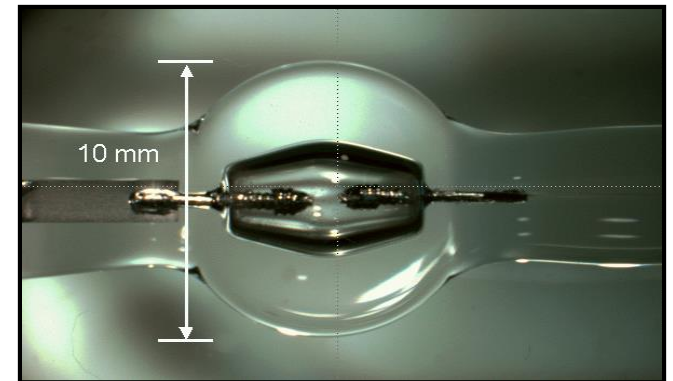
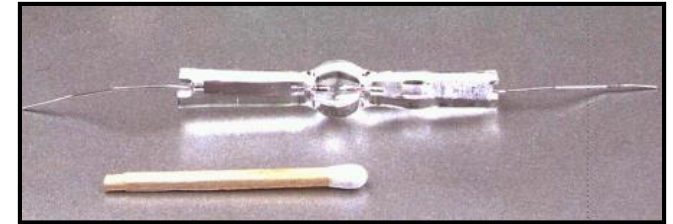
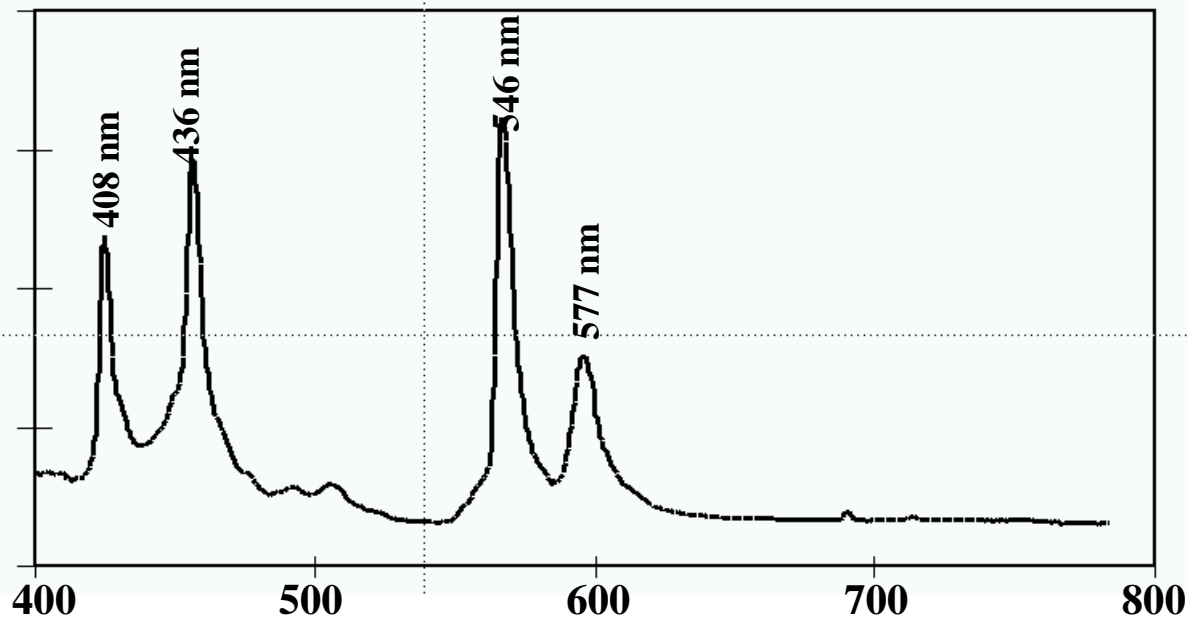
Requirements for light sources for projectors

- If possible punctual \Rightarrow A lot of light from a small volume
- High luminance (light density) \Rightarrow High Hg-pressure

UHP = **Ultra High Pressure (Performance)**

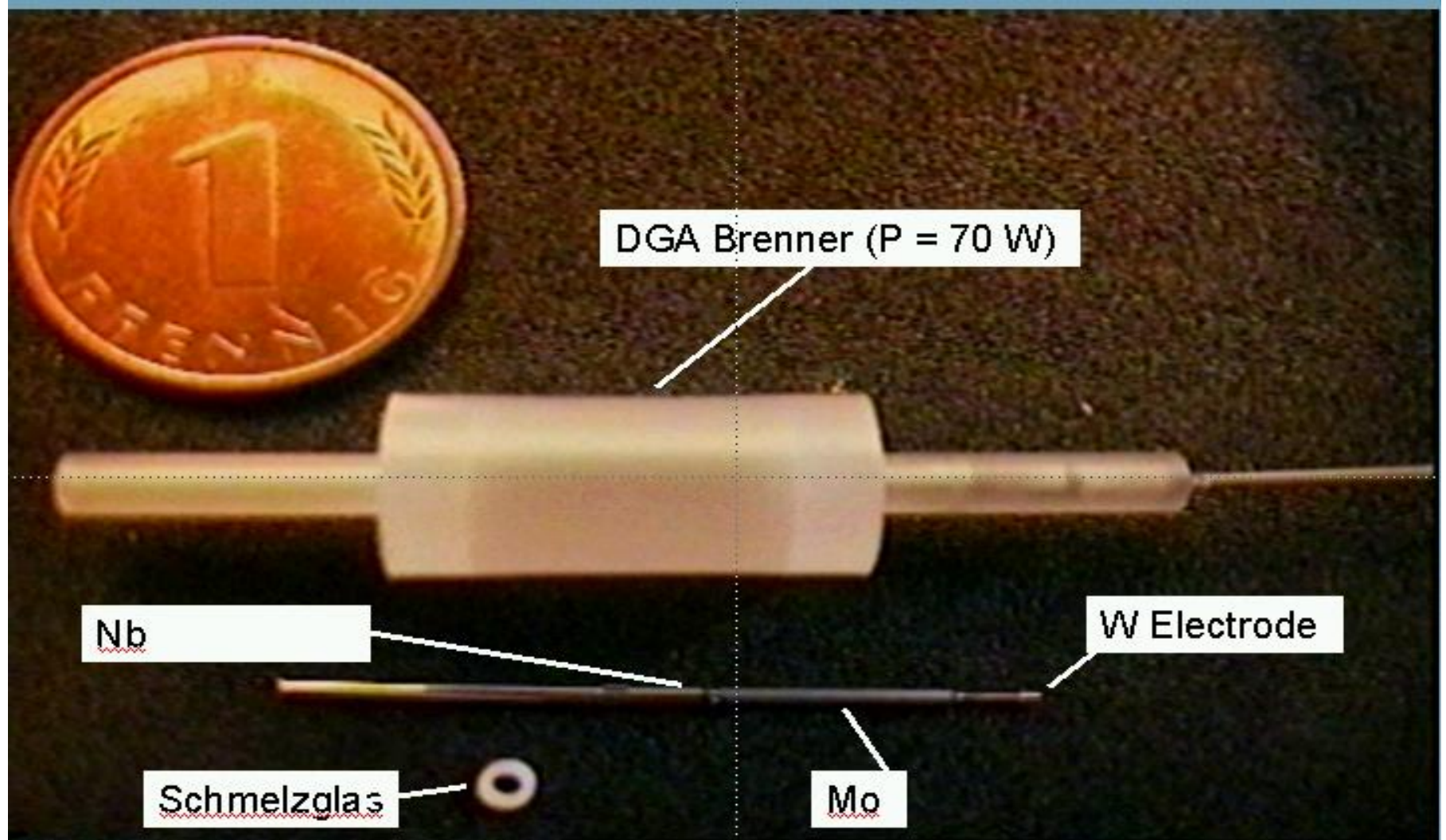
\Rightarrow Approx. 200 bar Hg, electrode separation ~ 1 mm

\Rightarrow Strong pressure-broadened lines of Hg



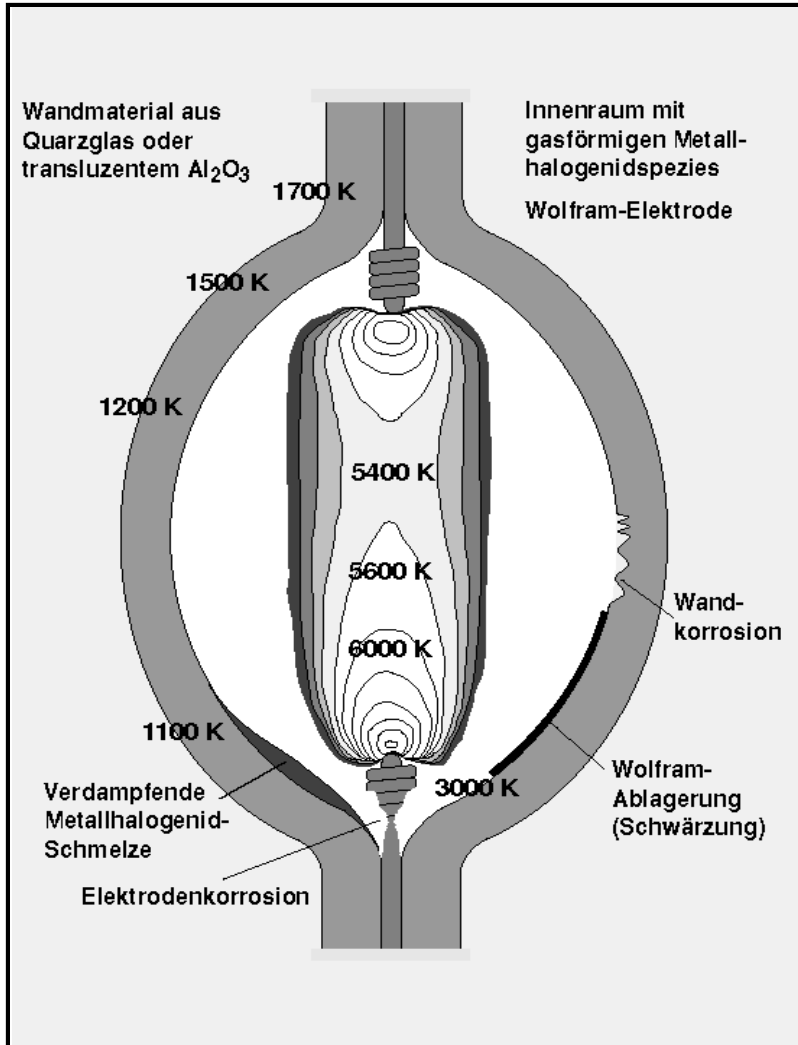
6.14 UHP-Lamps

Components of UHP-Lamps



6.14 UHP-Lamps

Design of UHP-lamps



Description of UHP-lamp by

- Chemical equations

Vapor pressure of metal halides

Disintegration of the metal halides in the plasma

- Temperature distribution in the plasma

Energy balance

Loss via radiation

Loss due to chemical energy

Loss due to heat

Convection (flow)

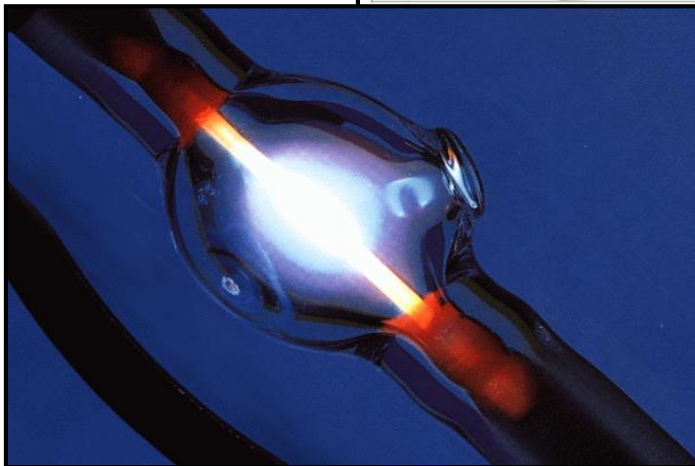
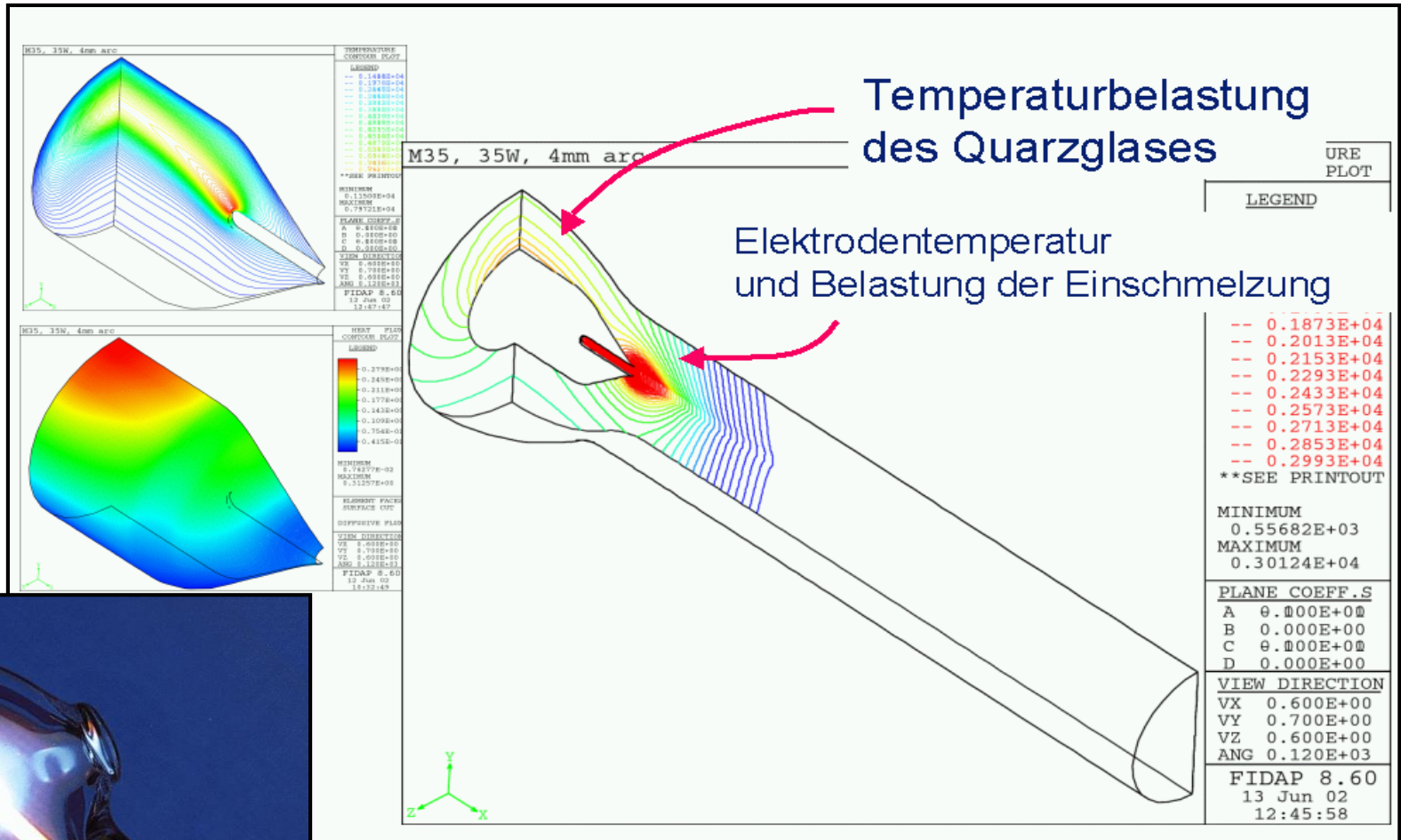
Heat conduction

- Convection equation = Navier-Stokes-Equation

$$\Rightarrow \frac{\partial^2 h}{\partial x'^2} + \frac{\partial^2 h}{\partial y'^2} = 0 \quad \text{Potential: } h = z + \frac{u}{\gamma w}$$

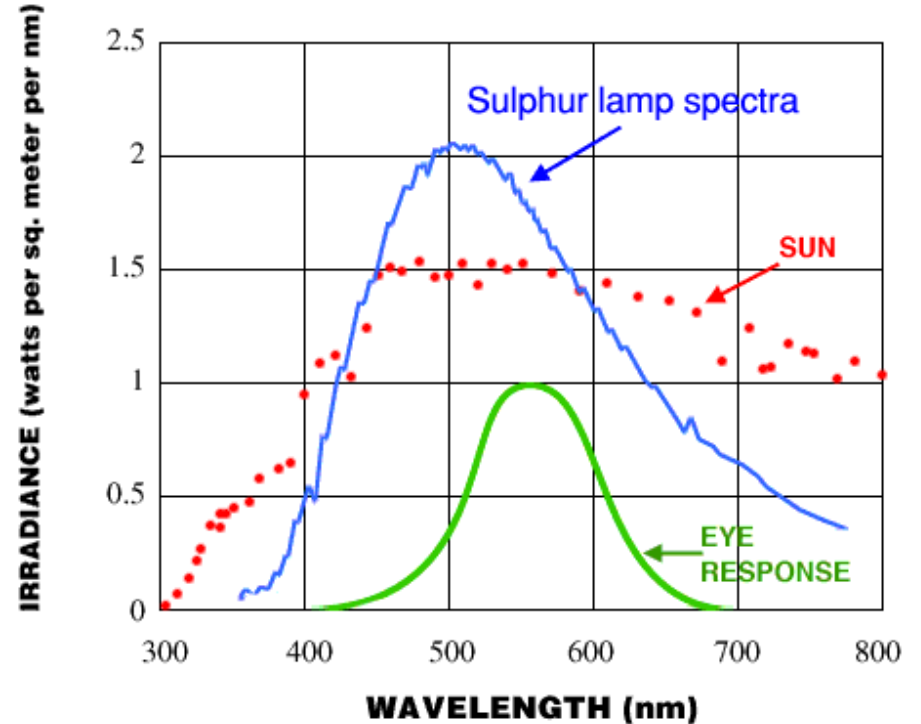
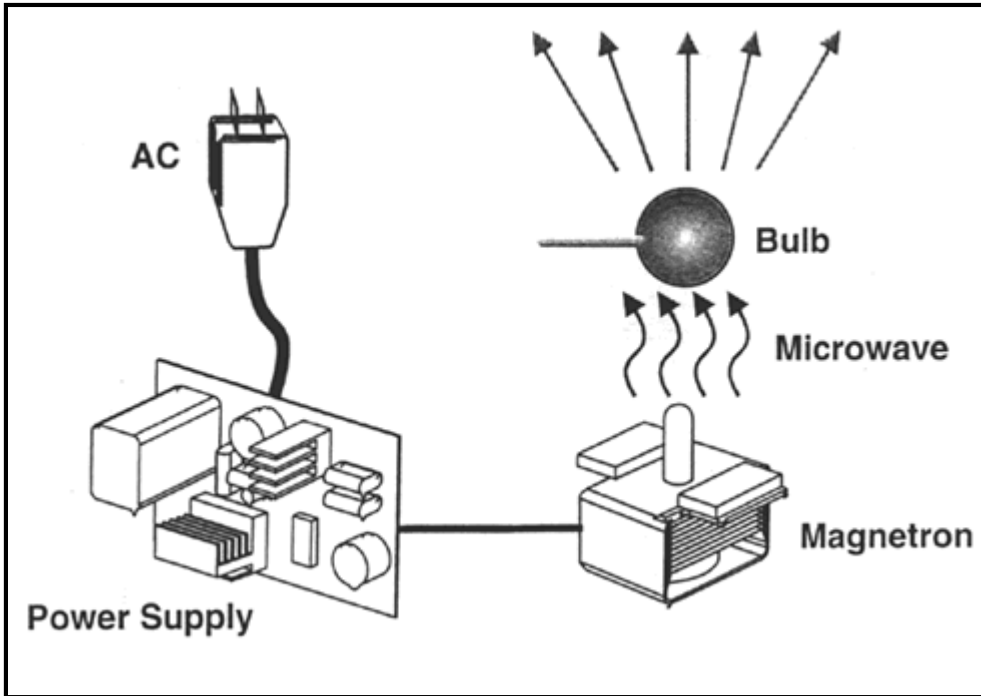
- Energy balance of the electrodes and the wall

6.14 UHP-Lamps



6.15 New Developments

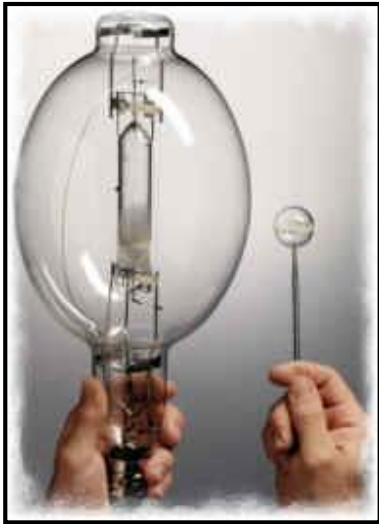
Sulfur lamp: In 1990 the first discharge lamp based on a molecular sulfur discharge ($S_4 - S_8$) was developed



The energy coupling into the discharge takes place by means of a microwave generator (magnetron), because electrodes can not be used

6.15 New Developments

Sulfur lamp: To generate a very large luminous flux



Typical operating parameters

Input power: 1.400 W

Ball diameter: approx. 30 mm

Luminous flux: 135000 lm

Color temperature: 5700 K

Starting time: 25 s

Lifetime (lamp): 60.000 h

Lifetime (magnetron): 20.000 h

Light output: 95 lm/W

Light source with extremely high light output, about 140000 lm (~ 40 fluorescent tubes) and (almost) pure-white light (emission band of S₈, ..., S₂ molecules)

Efficiency: Similar to fluorescent lights (thus 90 - 100 lm/W)

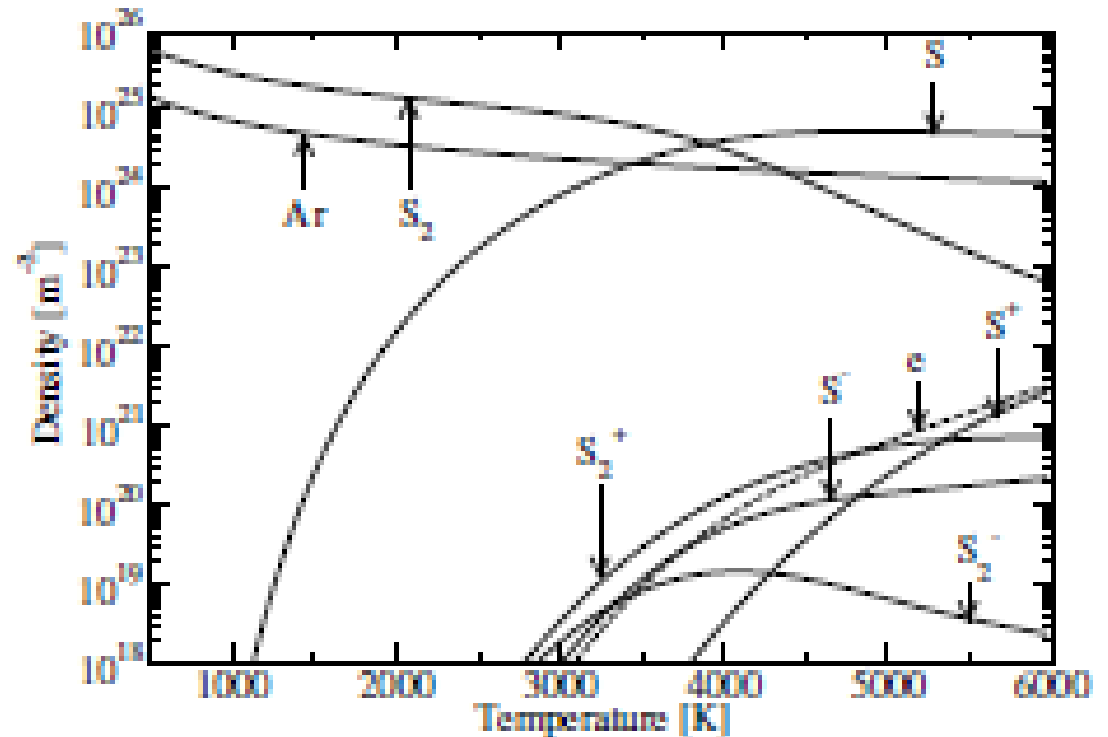
Problems: EMC and lifetime of the microwave generator

6.15 New Developments

Sulfur lamp: Mechanism of light generation \Rightarrow Emission from molecules, e.g. S_2

Reaction	Energy [eV]
$S_3 \rightleftharpoons S_3^+ + e$	10.6
$S_3^- \rightleftharpoons S_3 + e$	2.1
$S_3 \rightleftharpoons S_2 + S$	0.8
$S_2 \rightleftharpoons S_2^+ + e$	9.36
$S_2^- \rightleftharpoons S_2 + e$	1.67
$S_2 \rightleftharpoons S + S$	4.46
$S \rightleftharpoons S^+ + e$	10.36
$S^- \rightleftharpoons S + e$	2.1

Reactants	Products	ΔE [eV]
$S_2 + X$	$2S + X$	4.46
$S_2 + e$	$S_2^+ + e + e$	9.36
S_2^-	$S_2 + e$	1.8
$S + e$	$S^+ + e + e$	10.4
S^-	$S + e$	2.0
$Ar + e$	$Ar^+ + e + e$	15.76

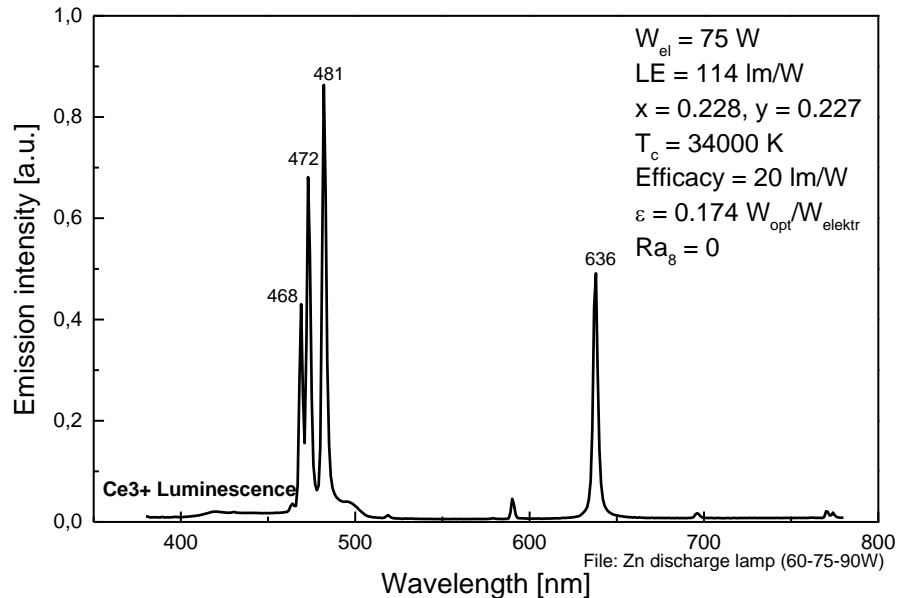


Lit.: C.W. Johnston, Transport and equilibrium in molecular plasmas: The sulfur lamp, Technische Universiteit Eindhoven, 2003

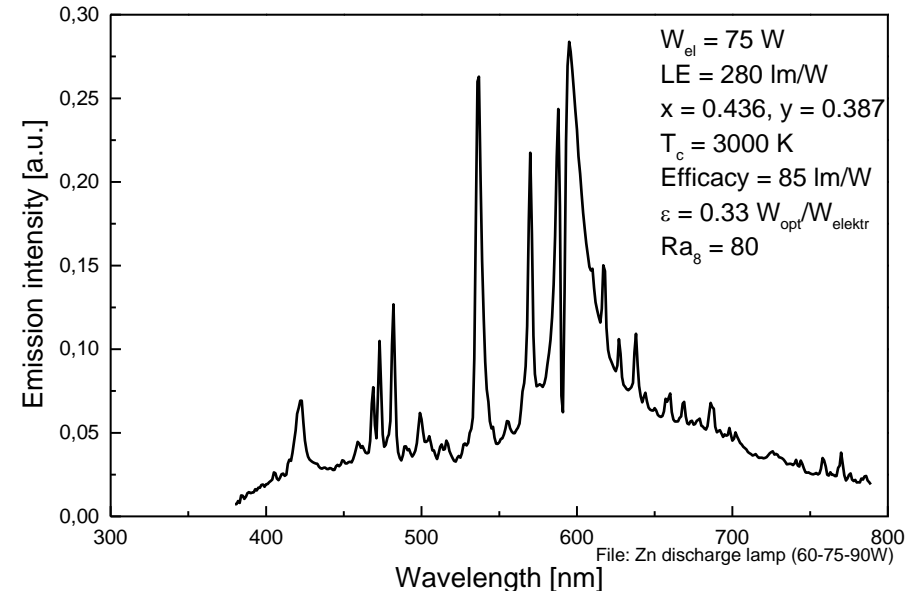
6.15 New Developments

Substitution of Hg by Zn (e.g. in automotive headlight lamps)

Zn/Ar Discharge



Zn/Ar/metal halide Discharge



Zn-Ar

Zn-Ar-metal halide

η 20 lm/W
Energy efficiency 17%
 R_a 0

85 lm/W
33%
80