

Examination

“Material Characterisation – Optical Spectroscopy (Prof. T. Jüstel)”

Date: February 02nd, 2012

Max. 25 Points

Name, Given name:

Enrolment number:

Please only use these sheets (you might also use the reverse)!

Task 1)

(5 Points)

Quantities and Terms

Please explain the following expressions!

- a) Radiometric quantities
- b) Photometric quantities
- c) Specular reflection
- d) Diffuse reflection
- e) Actinometry

Task 2)

(6 Points)

Luminescence spectroscopy

- a) Sketch the build-up of a typical fluorescence spectrometer and assign all required optical components!

- b) Describe the way to record an emission spectrum of a luminescent material, e.g. of $\text{Y}_2\text{O}_3:\text{Eu}^{3+}$ powder, that shows a charge-transfer transition at 230 nm!

- c) Describe the way to record an excitation spectrum of a luminescent material, e.g. of $\text{Y}_2\text{O}_3:\text{Eu}^{3+}$ powder, that shows an emission line at 611 nm!

- d) Why is it commonly necessary to correct excitation spectra? Please also describe the process of the correction!

Task 3)**(6 Points)****Reflection spectroscopy**

- a) Sketch the build-up of a typical reflection spectrometer and assign all required optical components!
- b) What is the function of the Ulbricht sphere?
- c) Please explain by taking the Kubelka-Munk-Function (R_∞ = reflectance, A = absorption coefficient und S = scattering coefficient) into account, why completely black substances do not exist!

Kubelka-Munk-Function:
$$F(R_\infty) = \frac{A}{S} = \frac{(1-R_\infty)^2}{2 \cdot R_\infty}$$

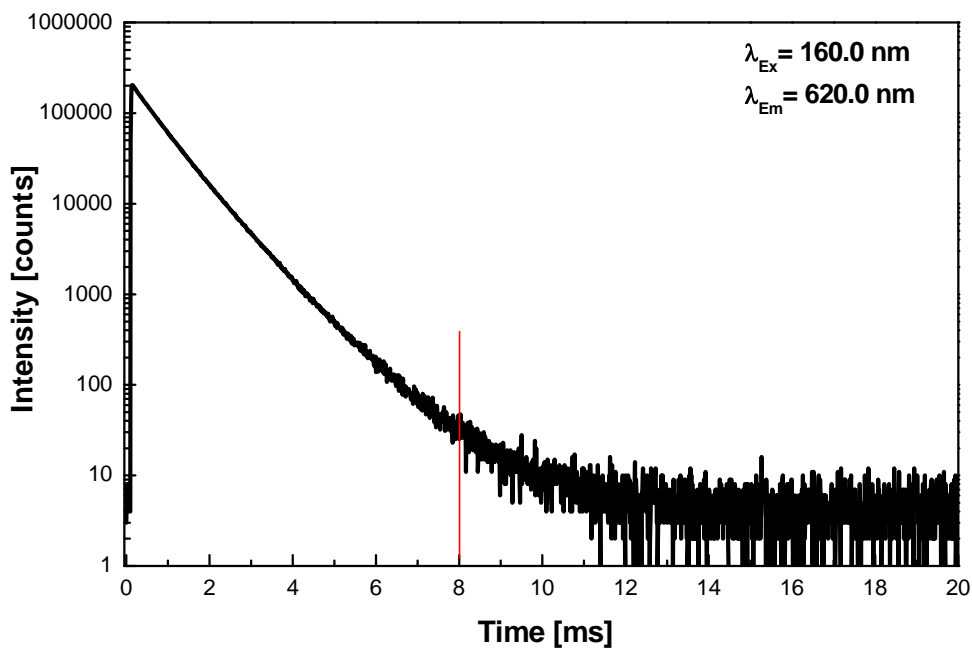
- d) Do completely white substances exist in accordance to this function? Please explain!

Task 4)

(4 Points)

Time resolved spectroscopy

- a) Describe the way to record a decay curve of a luminescent material,
- b) The figure below displays the decay curve of the high-pressure discharge lamp phosphor $\text{YVO}_4:\text{Eu}^{3+}$. Please determine the decay constants $\tau_{1/e}$ and $\tau_{1/10}$!



- c) Please name a potential cause for the deviation of the curve from linearity for the above given $\log(\text{Intensity})$ over time t plot about 5 ms after the excitation source has been switched off!

Task 5)**(4 Points)****Temperature resolved spectroscopy**

- a) Describe the way to record a thermal quenching curve and to determine the temperature $T_{1/2}$, i.e. the temperature, at which the luminescence intensity drops down to 50% relative to the low temperature luminescence intensity!
- b) Draw the shape of a typical thermal quenching curve in a respective diagram!