

# Making red emitting phosphors with $\text{Pr}^{3+}$

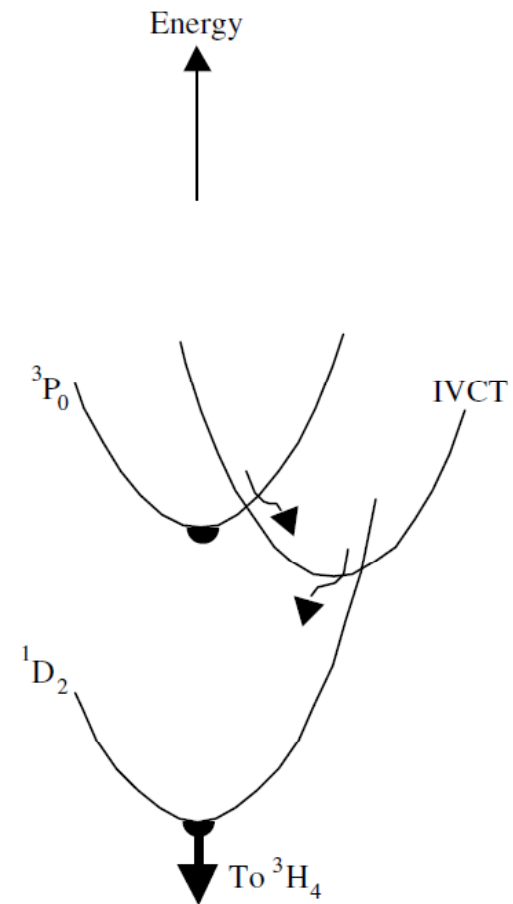
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# Basics

- $\text{Pr}^{3+}$  activated phosphors containing closed-shell transition metal ions show red luminescence
  - Titanates, vanadates, niobates
- UV excitation
- $^3\text{P}_0$  level (greenish-blue emission) quenched by intervalance charge transfer state (IVCT)
- $^1\text{D}_2 \rightarrow ^3\text{H}_4$ : red emission

# Quenching of $^3P_0$

- Intersystem crossing (f→d)
  - 4f5d band is too high in energy ( $60000\text{ cm}^{-1}$ )
- Cross relaxation
  - Limited due to doping  $< 0.2\text{ mol-}\%$
- Multiphonon relaxation
  - Only weak contribution (Dijk-Schuurman equation)
- IVCT
  - $\text{Pr}^{3+} + \text{M}^{n+} \rightarrow [\text{Pr}^{4+} + \text{M}^{(n-1)+}]$

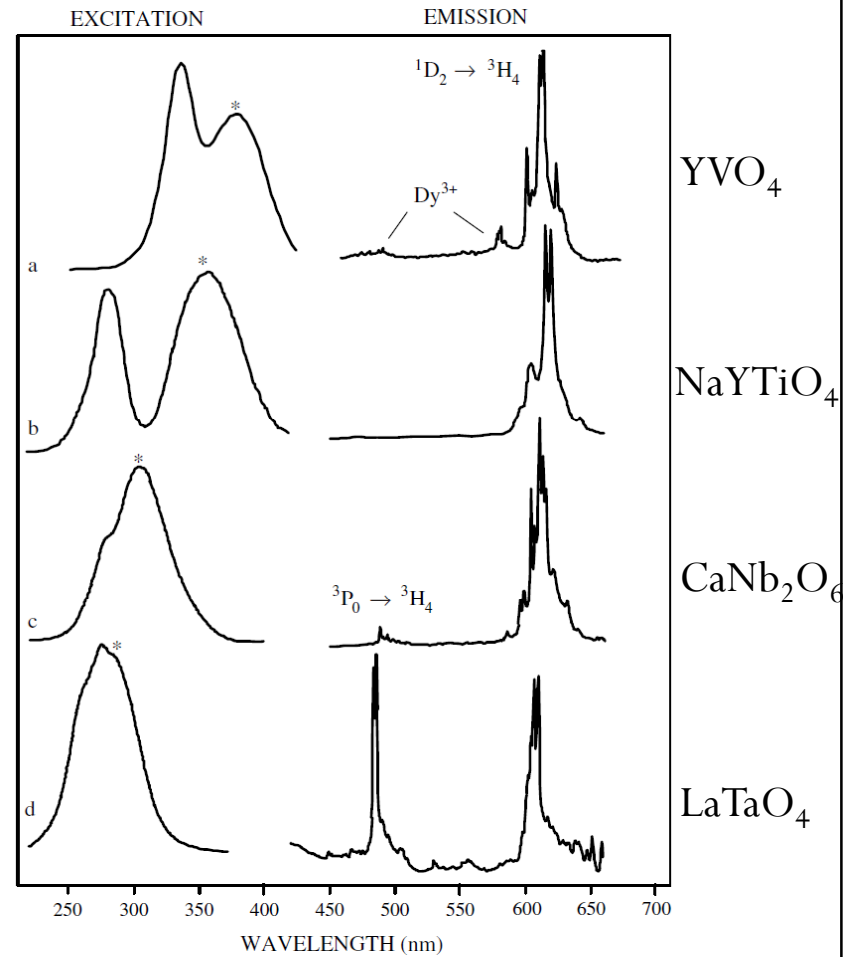


# Experimental

- Preparation of titanates, vanadates, niobates, tantalates
  - As crystalline powders by solid state reactions
  - As single crystals using the flux growth method
- Pr<sup>3+</sup> inserted in the rare earth or calcium sites
  - Only one site is available for the Pr<sup>3+</sup>

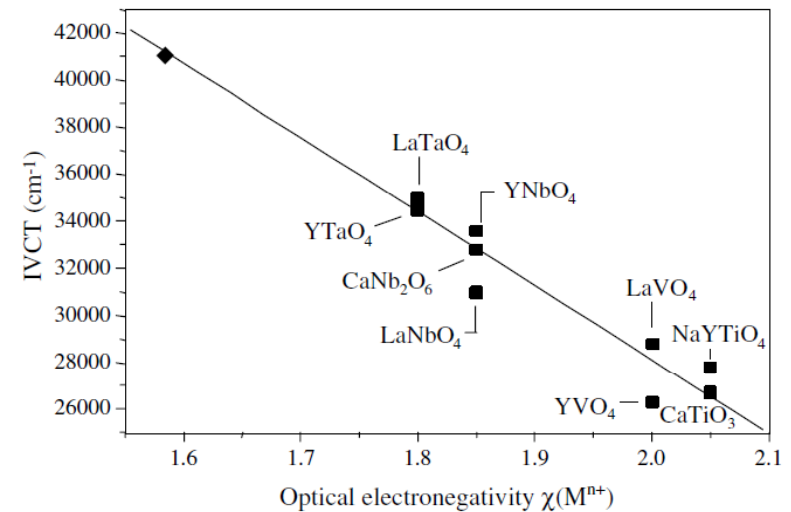
# Excitation and emission

- Two excitation bands
  - Host absorption (higher energy)
  - IVCT absorption (lower energy)



# IVCT and optical electronegativity

- Energetic position of the IVCT is roughly linear with the optical electronegativity
- $IVCT = 31\,450[2.89 - \chi(M^{n+})]$
- ${}^3P_0$ :  $\sim 20\,400\text{ cm}^{-1}$
- IVCT: Energy mismatch  $< 7400\text{ cm}^{-1}$



# Predicting $^3P_0$ quenching

Structural, vibrational and optical characteristics of closed-shell transition metal lattices containing  $\text{Pr}^{3+}$

Lattice	Avg(Pr-M) [ $\text{\AA}$ ]	$\hbar\omega_{\text{max}}$ [ $\text{cm}^{-1}$ ]	IVCT [ $\text{cm}^{-1}$ ]	Red/(red + blue)	$\chi(\text{M}^{n+})/\text{Avg}(\text{Pr-M})$
NaYTiO <sub>4</sub>	3.27	890	27,800	100%	0.627
CaTiO <sub>3</sub>	3.31	639	26,700	100%	0.619
YVO <sub>4</sub>	3.64	891	26,310	100%	0.550
LaVO <sub>4</sub>	3.71	860	28,800 <sup>a</sup>	>90%	0.540
CaNb <sub>2</sub> O <sub>6</sub>	3.67	904	32,800	>80%	0.504
YNbO <sub>4</sub>	3.73	830	33,600	>80%	0.496
YTaO <sub>4</sub>	3.72	825	34,480 <sup>a</sup>	>50%	0.484
LaNbO <sub>4</sub>	3.83	807	31,000 <sup>a</sup>	$\cong$ 50%	0.476
LaTaO <sub>4</sub>	3.81	810	35,000	$\cong$ 50%	0.472
CaZrO <sub>3</sub>	3.48	545	–	<20%	0.459

<sup>a</sup> The value is not accurate, Avg = average.

- Average distance (Avg(Pr-M)) between Pr and metal is also important
- Smaller distance leads to higher quenching rates
- Ratio of optical electronegativity and average distance is a simple criterion for predicting  $^3P_0$  quenching

# Conclusions

- Low-lying IVCT can be used to quench the  $^3P_0$  level
- Criterion: High ratio  $R = \chi(M^{n+}) / \text{Avg}(\text{Pr-M})$
- Red-emitting phosphors can be obtained by using the low cost  $\text{Pr}^{3+}$  ion
  - $\text{Pr}_2\text{O}_3$ : ~ 80 €/kg
  - $\text{Eu}_2\text{O}_3$ : ~ 1200 €/kg