

Verbindung	chemische Formel	$K_{sp}$ (25 °C)
Aluminiumhydroxid	$Al(OH)_3$	$3 \times 10^{-34}$
Aluminiumphosphat	$AlPO_4$	$9.84 \times 10^{-21}$
Aluminiumsulfid	$Al_2S_3$	$2.0 \times 10^{-7}$
Aluminiumselenid	$Al_2Se_3$	$4.0 \times 10^{-25}$
Bariumarsenat	$Ba_3(AsO_4)_2$	$8.0 \times 10^{-51}$
Bariumbromat	$Ba(BrO_3)_2$	$2.43 \times 10^{-4}$
Bariumcarbonat	$BaCO_3$	$2.58 \times 10^{-9}$
Bariumchromat	$BaCrO_4$	$1.17 \times 10^{-10}$
Bariumfluorid	$BaF_2$	$1.84 \times 10^{-7}$
Bariumhexafluorosilikat	$BaSiF_6$	$1.0 \times 10^{-6}$
Bariumhydroxidooctahydrat	$Ba(OH)_2 \times 8H_2O$	$2.55 \times 10^{-4}$
Bariumiodat	$Ba(IO_3)_2$	$4.01 \times 10^{-9}$
Bariumiodatmonohydrat	$Ba(IO_3)_2 \times H_2O$	$1.67 \times 10^{-9}$
Bariummolybdate	$BaMoO_4$	$3.54 \times 10^{-8}$
Bariumnitrat	$Ba(NO_3)_2$	$4.64 \times 10^{-3}$
Bariumpermanganat	$Ba(MnO_4)_2$	$2.5 \times 10^{-10}$
Bariumselenat	$BaSeO_4$	$3.40 \times 10^{-8}$
Bariumsulfat	$BaSO_4$	$1.08 \times 10^{-10}$
Bariumsulfid	$BaSO_3$	$5.0 \times 10^{-10}$
Berylliumcarbonattetrahydrat	$BeCO_3 \times 4H_2O$	$1 \times 10^{-3}$
Berylliumhydroxid	$Be(OH)_2$	$6.92 \times 10^{-22}$
Berylliummolybdat	$BeMoO_4$	$3.2 \times 10^{-2}$
Berylliumniobat	$Be(NbO_3)_2$	$1.2 \times 10^{-16}$
Bismutharsenat	$BiAsO_4$	$4.43 \times 10^{-10}$
Bismuthhydroxid	$Bi(OH)_3$	$4.0 \times 10^{-31}$
Bismuthiodid	$BiI_3$	$8.1 \times 10^{-19}$
Bismuthphosphat	$BiPO_4$	$1.3 \times 10^{-23}$
Blei(II)-bromid	$PbBr_2$	$6.60 \times 10^{-6}$
Blei(II)-carbonat	$PbCO_3$	$7.40 \times 10^{-14}$
Blei(II)-chlorid	$PbCl_2$	$1.70 \times 10^{-5}$
Blei(II)-chromat	$PbCrO_4$	$3 \times 10^{-13}$
Blei(II)-fluorid	$PbF_2$	$3.3 \times 10^{-8}$
Blei(II)-hydroxid	$Pb(OH)_2$	$1.43 \times 10^{-20}$
Blei(II)-iodat	$Pb(IO_3)_2$	$3.69 \times 10^{-13}$
Blei(II)-iodid	$PbI_2$	$9.8 \times 10^{-9}$
Blei(II)-oxalat	$PbC_2O_4$	$8.5 \times 10^{-9}$
Blei(II)-selenat	$PbSeO_4$	$1.37 \times 10^{-7}$
Blei(II)-sulfat	$PbSO_4$	$2.53 \times 10^{-8}$
Blei(II)-sulfid	$PbS$	$3 \times 10^{-28}$
Cadmiumarsenat	$Cd_3(AsO_4)_2$	$2.2 \times 10^{-33}$
Cadmiumcarbonat	$CdCO_3$	$1.0 \times 10^{-12}$
Cadmiumfluorid	$CdF_2$	$6.44 \times 10^{-3}$
Cadmiumhydroxid	$Cd(OH)_2$	$7.2 \times 10^{-15}$
Cadmiumiodat	$Cd(IO_3)_2$	$2.5 \times 10^{-8}$
Cadmiumoxalattrihydrat	$CdC_2O_4 \times 3H_2O$	$1.42 \times 10^{-8}$
Cadmiumphosphat	$Cd_3(PO_4)_2$	$2.53 \times 10^{-33}$

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Cadmiumsulfid	CdS	$1 \times 10^{-27}$
Cäsiumperchlorat	CsClO <sub>4</sub>	$3.95 \times 10^{-3}$
Cäsiumperiodat	CsIO <sub>4</sub>	$5.16 \times 10^{-6}$
Calciumcarbonat (Calcit)	CaCO <sub>3</sub>	$3.36 \times 10^{-9}$
Calciumcarbonat (Aragonit)	CaCO <sub>3</sub>	$6.0 \times 10^{-9}$
Calciumfluorid	CaF <sub>2</sub>	$3.45 \times 10^{-11}$
Calciumhydroxid	Ca(OH) <sub>2</sub>	$5.02 \times 10^{-6}$
Calciumiodat	Ca(IO <sub>3</sub> ) <sub>2</sub>	$6.47 \times 10^{-6}$
Calciumiodathexahydrat	Ca(IO <sub>3</sub> ) <sub>2</sub> × 6H <sub>2</sub> O	$7.10 \times 10^{-7}$
Calciummolybdat	CaMoO <sub>4</sub>	$1.46 \times 10^{-8}$
Calciumoxalatmonohydrat	CaC <sub>2</sub> O <sub>4</sub> × H <sub>2</sub> O	$2.32 \times 10^{-9}$
Calciumphosphat	Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	$2.07 \times 10^{-33}$
Calciumsulfat	CaSO <sub>4</sub>	$4.93 \times 10^{-5}$
Calciumsulfatdihydrat	CaSO <sub>4</sub> × 2H <sub>2</sub> O	$3.14 \times 10^{-5}$
Calciumsulfathemihydrat	CaSO <sub>4</sub> × 0.5H <sub>2</sub> O	$3.1 \times 10^{-7}$
Cobalt(II)-arsenat	Co <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub>	$6.80 \times 10^{-29}$
Cobalt(II)-carbonat	CoCO <sub>3</sub>	$1.0 \times 10^{-10}$
Cobalt(II)-hydroxid	Co(OH) <sub>2</sub>	$5.92 \times 10^{-15}$
Cobalt(II)-iodatdihydrat	Co(IO <sub>3</sub> ) <sub>2</sub> × 2H <sub>2</sub> O	$1.21 \times 10^{-2}$
Cobalt(II)-phosphat	Co <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	$2.05 \times 10^{-35}$
Cobalt(II)-sulfid (alpha)	CoS	$5 \times 10^{-22}$
Cobalt(II)-sulfid (beta)	CoS	$3 \times 10^{-26}$
Eisen(II)-carbonat	FeCO <sub>3</sub>	$3.13 \times 10^{-11}$
Eisen(II)-fluorid	FeF <sub>2</sub>	$2.36 \times 10^{-6}$
Eisen(II)-hydroxid	Fe(OH) <sub>2</sub>	$4.87 \times 10^{-17}$
Eisen(II)-sulfid	FeS	$8 \times 10^{-19}$
Eisen(III)-hydroxid	Fe(OH) <sub>3</sub>	$2.79 \times 10^{-39}$
Eisen(III)-phosphatdihydrat	FePO <sub>4</sub> × 2H <sub>2</sub> O	$9.91 \times 10^{-16}$
Europium(III)-hydroxid	Eu(OH) <sub>3</sub>	$9.38 \times 10^{-27}$
Gallium(III)-hydroxid	Ga(OH) <sub>3</sub>	$7.28 \times 10^{-36}$
Kaliumhexachloroplatinat	K <sub>2</sub> PtCl <sub>6</sub>	$7.48 \times 10^{-6}$
Kaliumperchlorat	KClO <sub>4</sub>	$1.05 \times 10^{-2}$
Kaliumperiodat	KIO <sub>4</sub>	$3.71 \times 10^{-4}$
Kupfer(I)-bromid	CuBr	$6.27 \times 10^{-9}$
Kupfer(I)-chlorid	CuCl	$1.72 \times 10^{-7}$
Kupfer(I)-cyanid	CuCN	$3.47 \times 10^{-20}$
Kupfer(I)-hydroxid	Cu <sub>2</sub> O	$2 \times 10^{-15}$
Kupfer(I)-iodid	CuI	$1.27 \times 10^{-12}$
Kupfer(I)-thiocyanat	CuSCN	$1.77 \times 10^{-13}$
Kupfer(II)-arsenate	Cu <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub>	$7.95 \times 10^{-36}$
Kupfer(II)-hydroxid	Cu(OH) <sub>2</sub>	$4.8 \times 10^{-20}$
Kupfer(II)-iodatmonohydrat	Cu(IO <sub>3</sub> ) <sub>2</sub> × H <sub>2</sub> O	$6.94 \times 10^{-8}$
Kupfer(II)-oxalat	CuC <sub>2</sub> O <sub>4</sub>	$4.43 \times 10^{-10}$
Kupfer(II)-phosphat	Cu <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	$1.40 \times 10^{-37}$
Kupfer(II)-sulfid	CuS	$8 \times 10^{-37}$
Lanthaniodat	La(IO <sub>3</sub> ) <sub>3</sub>	$7.50 \times 10^{-12}$
Lithiumcarbonat	Li <sub>2</sub> CO <sub>3</sub>	$8.15 \times 10^{-4}$

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Lithiumfluorid	LiF	$1.84 \times 10^{-3}$
Lithiumphosphat	$\text{Li}_3\text{PO}_4$	$2.37 \times 10^{-4}$
Magnesiumammoniumphosphat	$\text{MgNH}_4\text{PO}_4$	$3 \times 10^{-13}$
Magnesiumcarbonat	$\text{MgCO}_3$	$6.82 \times 10^{-6}$
Magnesiumcarbonattrihydrat	$\text{MgCO}_3 \times 3\text{H}_2\text{O}$	$2.38 \times 10^{-6}$
Magnesiumcarbonatpentahydrat	$\text{MgCO}_3 \times 5\text{H}_2\text{O}$	$3.79 \times 10^{-6}$
Magnesiumfluorid	$\text{MgF}_2$	$5.16 \times 10^{-11}$
Magnesiumhydroxid	$\text{Mg}(\text{OH})_2$	$5.61 \times 10^{-12}$
Magnesiumoxalatdihydrat	$\text{MgC}_2\text{O}_4 \times 2\text{H}_2\text{O}$	$4.83 \times 10^{-6}$
Magnesiumphosphat	$\text{Mg}_3(\text{PO}_4)_2$	$1.04 \times 10^{-24}$
Mangan(II)-carbonat	$\text{MnCO}_3$	$2.24 \times 10^{-11}$
Mangan(II)-hydroxid	$\text{Mn}(\text{OH})_2$	$2 \times 10^{-13}$
Mangan(II)-iodat	$\text{Mn}(\text{IO}_3)_2$	$4.37 \times 10^{-7}$
Mangan(II)-oxalatdihydrat	$\text{MnC}_2\text{O}_4 \times 2\text{H}_2\text{O}$	$1.70 \times 10^{-7}$
Mangan(II)-sulfid (pink)	MnS	$3 \times 10^{-11}$
Mangan(II)-sulfid (grün)	MnS	$3 \times 10^{-14}$
Neodym(III)-carbonat	$\text{Nd}_2(\text{CO}_3)_3$	$1.08 \times 10^{-33}$
Nickel(II)-carbonat	$\text{NiCO}_3$	$1.42 \times 10^{-7}$
Nickel(II)-hydroxid	$\text{Ni}(\text{OH})_2$	$5.48 \times 10^{-16}$
Nickel(II)-iodat	$\text{Ni}(\text{IO}_3)_2$	$4.71 \times 10^{-5}$
Nickel(II)-phosphat	$\text{Ni}_3(\text{PO}_4)_2$	$4.74 \times 10^{-32}$
Nickel(II)-sulfid (alpha)	NiS	$4 \times 10^{-20}$
Nickel(II)-sulfide (beta)	NiS	$1.3 \times 10^{-25}$
Palladium(II)-thiocyanat	$\text{Pd}(\text{SCN})_2$	$4.39 \times 10^{-23}$
Praseodym(III)-hydroxid	$\text{Pr}(\text{OH})_3$	$3.39 \times 10^{-24}$
Quecksilber(I)-bromid	$\text{Hg}_2\text{Br}_2$	$6.40 \times 10^{-23}$
Quecksilber(I)-carbonat	$\text{Hg}_2\text{CO}_3$	$3.6 \times 10^{-17}$
Quecksilber(I)-chlorid	$\text{Hg}_2\text{Cl}_2$	$1.43 \times 10^{-18}$
Quecksilber(I)-fluorid	$\text{Hg}_2\text{F}_2$	$3.10 \times 10^{-6}$
Quecksilber(I)-iodid	$\text{Hg}_2\text{I}_2$	$5.2 \times 10^{-29}$
Quecksilber(I)-oxalat	$\text{Hg}_2\text{C}_2\text{O}_4$	$1.75 \times 10^{-13}$
Quecksilber(I)-sulfat	$\text{Hg}_2\text{SO}_4$	$6.5 \times 10^{-7}$
Quecksilber(I)-thiocyanat	$\text{Hg}_2(\text{SCN})_2$	$3.2 \times 10^{-20}$
Quecksilber(II)-bromid	$\text{HgBr}_2$	$6.2 \times 10^{-20}$
Quecksilber(II)-hydroxid	HgO	$3.6 \times 10^{-26}$
Quecksilber(II)-iodid	$\text{HgI}_2$	$2.9 \times 10^{-29}$
Quecksilber(II)-sulfid (schwarz)	HgS	$2 \times 10^{-53}$
Quecksilber(II)-sulfid (rot)	HgS	$2 \times 10^{-54}$
Radiumiodat	$\text{Ra}(\text{IO}_3)_2$	$1.16 \times 10^{-9}$
Radiumsulfat	$\text{RaSO}_4$	$3.66 \times 10^{-11}$
Rubidiumperchlorat	$\text{RuClO}_4$	$3.00 \times 10^{-3}$
Scandiumfluorid	$\text{ScF}_3$	$5.81 \times 10^{-24}$
Scandiumhydroxid	$\text{Sc}(\text{OH})_3$	$2.22 \times 10^{-31}$
Silber(I)-acetat	$\text{AgCH}_3\text{COO}$	$1.94 \times 10^{-3}$
Silber(I)-arsenat	$\text{Ag}_3\text{AsO}_4$	$1.03 \times 10^{-22}$
Silber(I)-bromat	$\text{AgBrO}_3$	$5.38 \times 10^{-5}$
Silber(I)-bromid	AgBr	$5.35 \times 10^{-13}$
Silber(I)-carbonat	$\text{Ag}_2\text{CO}_3$	$8.46 \times 10^{-12}$

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Silber(I)-chlorid	AgCl	$1.77 \times 10^{-10}$
Silber(I)-chromat	Ag <sub>2</sub> CrO <sub>4</sub>	$1.12 \times 10^{-12}$
Silber(I)-cyanid	AgCN	$5.97 \times 10^{-17}$
Silber(I)-iodat	AgIO <sub>3</sub>	$3.17 \times 10^{-8}$
Silber(I)-iodid	AgI	$8.52 \times 10^{-17}$
Silber(I)-oxalat	Ag <sub>2</sub> C <sub>2</sub> O <sub>4</sub>	$5.40 \times 10^{-12}$
Silber(I)-phosphat	Ag <sub>3</sub> PO <sub>4</sub>	$8.89 \times 10^{-17}$
Silber(I)-sulfat	Ag <sub>2</sub> SO <sub>4</sub>	$1.20 \times 10^{-5}$
Silber(I)-sulfid	Ag <sub>2</sub> SO <sub>3</sub>	$1.50 \times 10^{-14}$
Silber(I)-sulfid	Ag <sub>2</sub> S	$8 \times 10^{-51}$
Silber(I)-thiocyanat	AgSCN	$1.03 \times 10^{-12}$
Strontiumarsenat	Sr <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub>	$4.29 \times 10^{-19}$
Strontiumcarbonat	SrCO <sub>3</sub>	$5.60 \times 10^{-10}$
Strontiumfluorid	SrF <sub>2</sub>	$4.33 \times 10^{-9}$
Strontiumiodat	Sr(IO <sub>3</sub> ) <sub>2</sub>	$1.14 \times 10^{-7}$
Strontiumiodatmonohydrat	Sr(IO <sub>3</sub> ) <sub>2</sub> × H <sub>2</sub> O	$3.77 \times 10^{-7}$
Strontiumiodathexahydrat	Sr(IO <sub>3</sub> ) <sub>2</sub> × 6H <sub>2</sub> O	$4.55 \times 10^{-7}$
Strontiumoxalat	SrC <sub>2</sub> O <sub>4</sub>	$5 \times 10^{-8}$
Strontiumsulfat	SrSO <sub>4</sub>	$3.44 \times 10^{-7}$
Thallium(I)-bromat	TlBrO <sub>3</sub>	$1.10 \times 10^{-4}$
Thallium(I)-bromid	TlBr	$3.71 \times 10^{-6}$
Thallium(I)-chlorid	TlCl	$1.86 \times 10^{-4}$
Thallium(I)-chromat	Tl <sub>2</sub> CrO <sub>4</sub>	$8.67 \times 10^{-13}$
Thallium(I)-hydroxid	Tl(OH) <sub>3</sub>	$1.68 \times 10^{-44}$
Thallium(I)-iodat	TlIO <sub>3</sub>	$3.12 \times 10^{-6}$
Thallium(I)-iodid	TlI	$5.54 \times 10^{-8}$
Thallium(I)-thiocyanat	TlSCN	$1.57 \times 10^{-4}$
Thallium(I)-sulfid	Tl <sub>2</sub> S	$6 \times 10^{-22}$
Thuliumhydroxid	Tm(OH) <sub>3</sub>	$3.3 \times 10^{-24}$
Titan(III)-hydroxid	Ti(OH) <sub>3</sub>	$1.0 \times 10^{-40}$
Titanylhydroxid	TiO(OH) <sub>2</sub>	$1.0 \times 10^{-29}$
Vanadyl(IV)-hydroxid	VO(OH) <sub>2</sub>	$5.9 \times 10^{-23}$
Vanadyl(III)-phosphat	(VO) <sub>3</sub> PO <sub>4</sub>	$8 \times 10^{-25}$
Ytterbiumhydroxid	Yb(OH) <sub>3</sub>	$2.5 \times 10^{-24}$
Yttriumcarbonat	Y <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	$1.03 \times 10^{-31}$
Yttriumfluorid	YF <sub>3</sub>	$8.62 \times 10^{-21}$
Yttriumhydroxid	Y(OH) <sub>3</sub>	$1.0 \times 10^{-22}$
Yttriumiodat	Y(IO <sub>3</sub> ) <sub>3</sub>	$1.12 \times 10^{-10}$
Zinkarsenat	Zn <sub>3</sub> (AsO <sub>4</sub> ) <sub>2</sub>	$2.8 \times 10^{-28}$
Zinkcarbonat	ZnCO <sub>3</sub>	$1.46 \times 10^{-10}$
Zinkcarbonatmonohydrat	ZnCO <sub>3</sub> × H <sub>2</sub> O	$5.42 \times 10^{-11}$
Zinkfluorid	ZnF <sub>2</sub>	$3.04 \times 10^{-2}$
Zinkhydroxid	Zn(OH) <sub>2</sub>	$3.0 \times 10^{-17}$
Zinkiodatdihydrat	Zn(IO <sub>3</sub> ) <sub>2</sub> × 2H <sub>2</sub> O	$4.1 \times 10^{-6}$
Zinkmetaboratmonohydrat	Zn(BO <sub>2</sub> ) <sub>2</sub> × H <sub>2</sub> O	$6.6 \times 10^{-11}$
Zinkoxalatdihydrat	ZnC <sub>2</sub> O <sub>4</sub> × 2H <sub>2</sub> O	$1.38 \times 10^{-9}$
Zinkphosphat	Zn <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	$9.0 \times 10^{-33}$

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Zinkselenid	ZnSe	$3.6 \times 10^{-26}$
Zinkselenitmonohydrat	ZnSeO <sub>3</sub> ×H <sub>2</sub> O	$1.59 \times 10^{-7}$
Zinksulfid (alpha)	ZnS	$2 \times 10^{-25}$
Zinksulfid (beta)	ZnS	$3 \times 10^{-23}$
Zinn(II)-hydroxid	Sn(OH) <sub>2</sub>	$5.45 \times 10^{-27}$
Zinn(IV)-hydroxid	Sn(OH) <sub>4</sub>	$1.0 \times 10^{-56}$
Zinn(II)-sulfid	SnS	$1.0 \times 10^{-25}$
Zirconiumoxidhydroxid	ZrO(OH) <sub>2</sub>	$6.3 \times 10^{-49}$
Zirconiumphosphat	Zr <sub>3</sub> (PO <sub>4</sub> ) <sub>4</sub>	$1.0 \times 10^{-132}$